## Cosmic Evolution

# Part 1: Protons to heavy elements

Big Bang occurred 13.7 Billion yrs ago (13.7 x 10<sup>9</sup> yr) Only fundamental particles existed for first few minutes

Name	Symbol	Charge	Mass
Proton	р	+	1.7 × 10 <sup>-24</sup> g
Neutron	n	0	1.7 × 10 <sup>-24</sup> g
Electron	е	-	1 × 10 <sup>-27</sup> g
Photon	γ	0	0
Neutrino	ν	0	< 10 <sup>-33</sup> g

Building blocks of nuclei but only one kind of nucleus

Proton = nucleus of Hydrogen

# The Bigger Picture

The ordinary matter (protons, neutrons, ...) contain only 4.5% of the mass energy of the Universe. Dark matter contains 22.7% and the even stranger dark energy accounts for 72.8%. (most recent numbers)

# A Bit of Physics

Energy of Motion (Kinetic Energy)

$$E = \frac{1}{2} \text{ mv}^2 \text{ (if v not close to c)}$$
Gas at Temperature T,
$$Avg. \text{ Energy} \qquad E = \frac{3}{2} \text{ kT}$$

Avg. Energy 
$$E = {}^{3}kT$$

So avg. v : 
$$\frac{1}{2} \text{ mv}^2 = \frac{3}{2} \text{ kT}$$

$$V = \sqrt{\frac{3kT}{m}} = \frac{3kT}{2}$$

Higher  $T \rightarrow$  Higher v, E on avg.

## Simulation of gas properties on the web

 $\underline{\text{http://phet.colorado.edu/new/simulations/sims.php?sim=Gas\_Properties}}$ 

Early Universe so hot that collisions broke apart any complex things that might have formed

As Universe expanded, T dropped

at  $\sim 3$  min, T  $\sim 10^9$  K

A few nuclei form (nucleosynthesis)

at  $\sim 30$  min,  $T \sim 3 \times 10^8 \ K~$  end of nucleosynthesis

### Composition of Universe at 30 min.

~ 94% proton

 $\sim 6\%$  alpha particle He

(and electrons)

At 380,000 years T ~ 3000 K

Nuclei + electron  $\rightarrow$  Atoms H He

## First Generation Stars

Expanding Universe

But, Gravity collected matter into Stars

Stars now 0.1 to 100  ${\rm M}_{\odot}$ ; first stars more massive

Later into Galaxies  $\,(M\sim 10^{10}\,\text{to}\,10^{12}\,M_\odot)\,$ 

Oldest stars in disk: age  $\sim 10^{10}$  years

First generation stars  $\rightarrow$  No C, O, N, ...

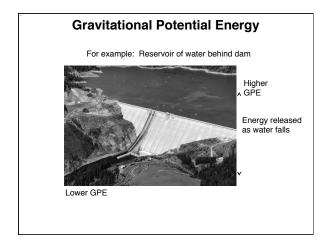
 $\Rightarrow$  No life No Si, Fe  $\Rightarrow$  No Earthlike planets

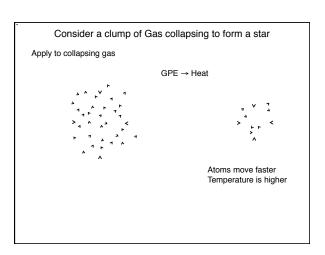
But they made some "heavy" elements

So later stars could have solid planets, life

Movie illustrating galaxy formation

From http://cosmicweb.uchicago.edu/group.html





## Pause for Demonstration

# Back to Formation of First Stars Collapse released Gravitational Potential Energy The gas heats up The Temperature in core reaches 10<sup>7</sup> K Nuclear reactions begin Collapse stops Why?

## **Nuclear Potential Energy**

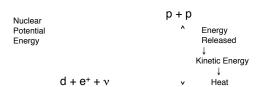
Four basic forces: gravity, electromagnetic, weak and strong nuclear force

Each has potential energy. Nuclear potential energy can be released by nuclear reactions.

e.g. 1st step:  $p + p \rightarrow d + e^+ + v$ 

d = deuteron = proton + neutron e<sup>+</sup> = positron (antiparticle of electron)

# **Nuclear Potential Energy**



Separation of two protons

The energy released by nuclear reactions supplies heat → pressure

Resists gravity ⇒ stable star

## **Electromagnetic Barrier**

Why do we need high T ( $\sim 10^7$  K)?

Protons have positive electric charge Like Charges Repel

As protons approach, repulsion grows, corresponds to climbing hill of electromagnetic potential energy

# **Electromagnetic Barrier**

Barrier potential Potenty

 $d + e^+ + v$ 

Separation

Barrier is really much higher than  $\frac{3}{2}~k\cdot 10^7~K$  Very few can get over barrier

 $\Rightarrow$  Stars live a long time rather than exploding



## Questions

- Why do nuclear reactions produce a longlived system in star, but an explosion in a bomb?
- What will happen when a star's fuel runs out?

```
proton p
neutron n
positron e+
neutrino v
photon \gamma

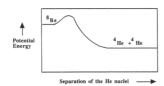
p + p \rightarrow d + e^+ + v_{\tau}
\rightarrow \qquad \qquad \Rightarrow
d + p \rightarrow {}^{3}He + {}^{7}
\rightarrow \qquad \qquad \Rightarrow
{}^{3}He + {}^{3}He \rightarrow {}^{4}He + 2p
\rightarrow \qquad \qquad \Rightarrow
4

Nucleosynthesis
Again

Nucleosynthesis
Again
```

How to get past helium? We need C, O, N, P, S, ...

 $^4\text{He} + ^4\text{He} \rightarrow ^8\text{Be}$   $^8\text{Be} = 4\text{p} + 4\text{n}$  Problem:  $^8\text{Be}$  has more nuclear potential energy than parts; It is unstable (radioactive).



To get carbon, we need another  $^4\mbox{He}$  to hit  $^8\mbox{Be}$  before  $^8\mbox{Be}$  falls apart

$$^{4}$$
He +  $^{4}$ He →  $^{8}$ Be  
→  $^{8}$ Be =  $^{4}$ P +  $^{4}$ Π  
 $^{8}$ Be +  $^{4}$ He →  $^{12}$ C +  $^{7}$   
→  $^{12}$ C =  $^{6}$ P +  $^{6}$ Π  
 $^{4}$ He +  $^{12}$ C →  $^{16}$ Ω  $^{16}$ Ω =  $^{8}$ P +  $^{8}$ Π  
 $^{16}$ Ω +  $^{16}$ Ω →  $^{32}$ S +  $^{7}$ Λ Sulfur  
 $^{16}$ Ω +  $^{16}$ Ω →  $^{31}$ P +  $^{9}$ Λ Phosphorus  
 $^{16}$ Ω +  $^{16}$ Ω →  $^{28}$ Si +  $^{4}$ He Silicon

## Questions

- · What was needed to make the bioelements?
- · Are any missing?
- How do the bioelements get out of the star?



# **Summary**

Heavy elements needed for life were created by early generations of massive stars.

Except for H, we are made of star debris

Natural forces (Gravity, EM, Nuclear) produced first evolution of matter from simple to complex (protons → heavy elements)