

2/1/07

Physics of Compact Objects (cont)

Pressure inside stars:

know: n, T, ρ , etc. \Rightarrow Calculate $P = P(n, T, \dots)$
Equation of state \nearrow

Possible combinations of conditions of gas:

classical (non-degenerate)	quantum ("degenerate")
non-relativistic (NR)	ultra-relativistic (UR)
$E_{kin} \ll m_0 c^2$	$E_{kin} \gg m_0 c^2$

Case 1: classical + NR ("normal" \rightarrow atmosphere of Earth)

gas is NR $\Rightarrow P = \frac{2}{3} U_{kin}$

$$U_{kin} = \frac{N}{V} \langle E_{kin} \rangle = n \langle E_{kin} \rangle$$

$$= \frac{3}{2} n k_B T$$

$\left(\frac{1}{2} kT \right.$ per degree of freedom per particle $\left. \right)$

\nearrow Boltzmann constant

$$\Rightarrow p = \frac{2}{3} U_{\text{kin}} = n k_B T \quad (\text{"ideal gas law"})$$

Classical/Quantum Boundary

Q: When does quantum mechanics (QM) become important?

Define the average distance between particles:

$$n = \frac{N}{V} = \frac{1}{l^3} \Rightarrow l = n^{-1/3},$$

where l is the average distance between particles

Consider the QM wave associated with a particle:

$$\lambda_{\text{deB}} = \frac{h}{p}, \quad \text{where } h \text{ is Planck's constant.}$$

QM important when $l \leq \lambda_{\text{deB}}$.

$$\Rightarrow n^{-1/3} \leq \frac{h}{p} \Rightarrow p \leq h n^{1/3}$$

\Rightarrow Slow-moving particles, which are dense, can become quantum gases.

Now, slow-moving \Rightarrow cold, as

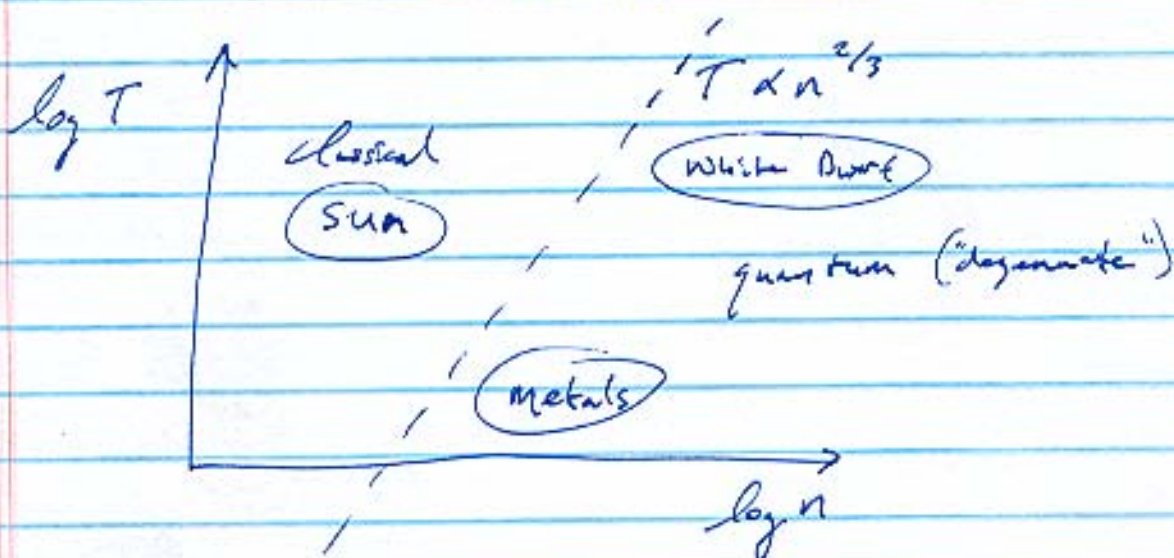
$$\frac{3}{2} k_B T = \frac{p^2}{2m_0}$$

$$\Rightarrow k_B T \approx \frac{p^2}{m_0} \leq \frac{h^2 n^{2/3}}{m_0}$$

So, the classical / quantum boundary exists along

$$T \propto n^{2/3}, \text{ or}$$

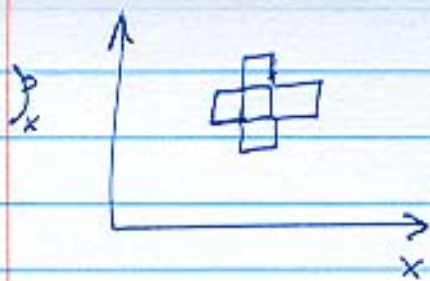
$$T = \text{constant} \times n^{2/3}$$



Case 2: Quantum gas

Consider phase-space (6-dimensional)
 \downarrow
(x, y, z, p_x, p_y, p_z)

Look at a slice of phase space:



quantum cells

$$\underbrace{dx dy dz}_{dV} dp_x dp_y dp_z = h^3$$

From Heisenberg's
Uncertainty Principle: $dx dp_x \gtrsim h$