Cosmic Evolution

Part 1: Protons to heavy elements

Big Bang occurred 13.7 Billion yrs ago (13.7 x 10⁹ yr)
Only fundamental particles existed for first few minutes

<u>Name</u>	<u>Symbol</u>	<u>Charge</u>	<u>Mass</u>
Danta			4.7. 4.0 –24 a.
Proton	р	+	$1.7 \times 10^{-24} \text{ g}$
Neutron	n	0	$1.7 \times 10^{-24} \mathrm{g}$
Electron	е	_	$1 \times 10^{-27} \mathrm{g}$
Photon	γ	0	0
Neutrino	ν	0	~ 0 (?)

Building blocks of nuclei but only one kind of nucleus Proton = nucleus of Hydrogen

Energy of Motion (Kinetic Energy)

$$E = \frac{1}{2} \text{ mv}^2$$
 (if v not close to c)

Gas at Temperature T,

Avg. Energy
$$E = \frac{3}{2} kT$$

So avg. v :
$$\frac{1}{2} \text{ mv}^2 = \frac{3}{2} \text{ kT}$$

$$V = \sqrt{\frac{3kT}{m}} = \left(\frac{3kT}{m}\right)^{\frac{1}{2}}$$

Higher $T \rightarrow Higher v$, E on avg.

Early Universe so hot that collisions broke apart any complex things that might have formed

As Universe expanded, T dropped at ~ 3 min, T $\sim 10^9$ K A few nuclei form (nucleosynthesis) at ~ 30 min, T $\sim 3 \times 10^8$ K end of nucleosynthesis Composition of Universe at 30 min.

~ 94 % proton ~ 6% alpha part. (and electrons)

At 380,000 years T ~ 3000

Nuclei + electron → Atoms

First Generation Stars

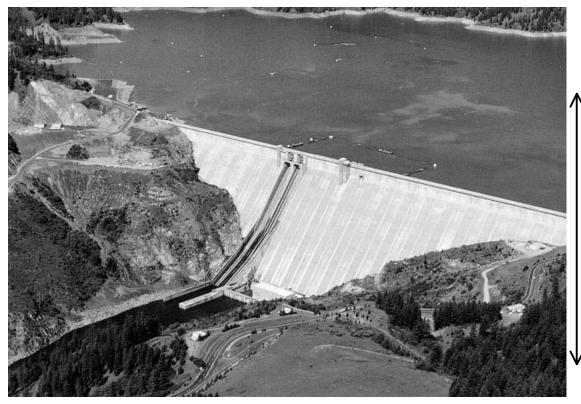
Expanding Universe But, Gravity \rightarrow Galaxies ($\sim 10^{11}$ to 10^{12} M $_{\odot}$) \rightarrow Stars (0.1 to 100 M $_{\odot}$)

Oldest stars in disk ~ 10¹⁰ y old

First generation stars \rightarrow No C, O, N, ... \Rightarrow No life No Si, Fe \Rightarrow No Earthlike planets But they made "heavy" elements So later stars could have solid planets, life

Gravitational Potential Energy

For example: Reservoir of water behind dam

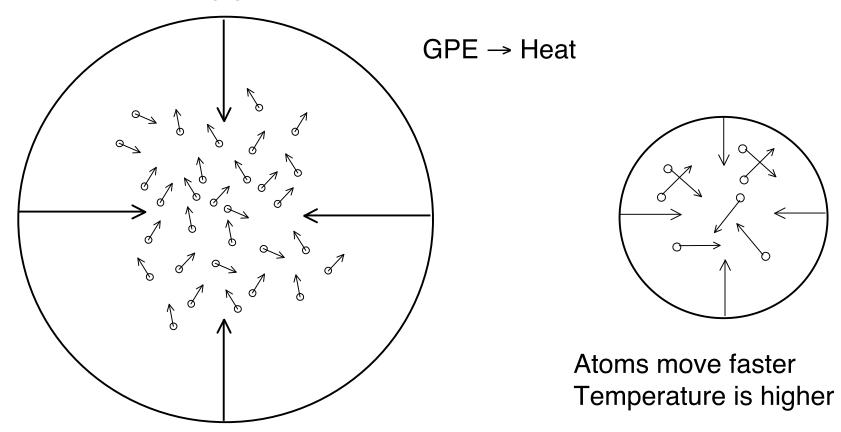


Energy released as water falls

Lower GPE

Consider a clump of Gas collapsing to form a star

Apply to collapsing gas



Pause for Demonstration

Back to Formation of First Stars

Collapse released Gravitational Potential Energy

The gas heats up

The Temperature in core reaches 10⁷ k

Nuclear reactions begin

Collapse stops

Why?

Nuclear Potential Energy

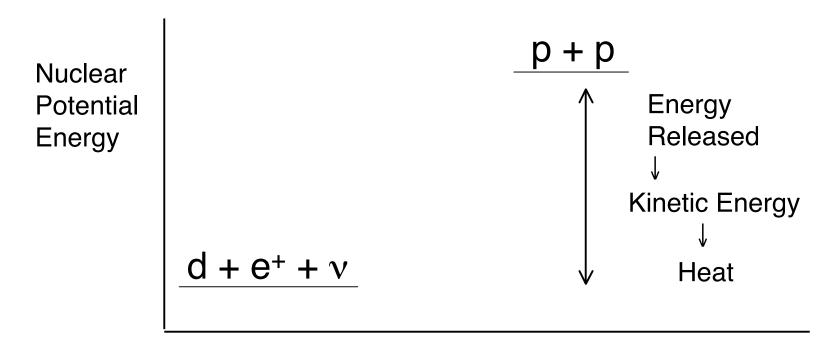
Four basic forces: gravity, electromagnetic, weak and strong nuclear force

Each has potential energy. Nuclear potential energy can be released by nuclear reactions.

e.g. 1st step:
$$p + p \rightarrow d + e^+ + v$$

d = deuteron = proton + neutron e⁺ = positron (antiparticle of electron)

Nuclear Potential Energy



Separation of two protons

The energy released by nuclear reactions supplies heat → pressure

Resists gravity ⇒ stable star

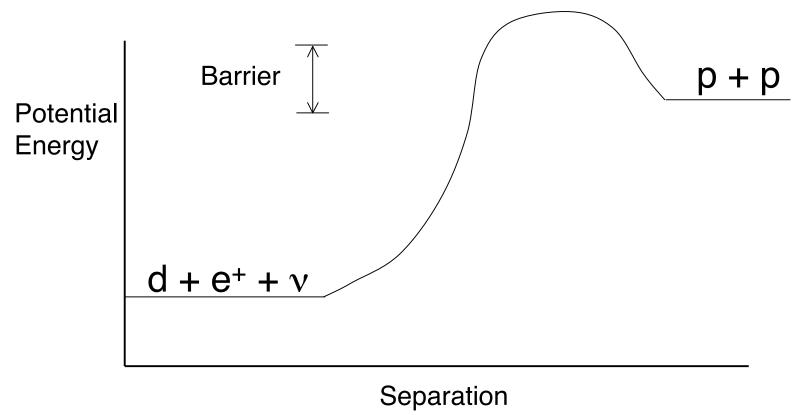
Electromagnetic Barrier

Why do we need high T ($\sim 10^7$ K)?

Protons have positive electric charge Like Charges Repel

As protons approach, repulsion grows, corresponds to climbing hill of electromagnetic potential energy

Electromagnetic Barrier



Barrier is really much higher than $\frac{3}{2} k \cdot 10^7 k$ Very few can get over barrier

⇒ Stars live a long time rather than exploding



Questions

- Why do nuclear reactions produce a longlived system in star, but an explosion in a bomb?
- What will happen when a star's fuel runs out?

n

$$\sim$$
 neutrino \sim photon

$$p + p \rightarrow d + e^{+} + v$$

$$0 \rightarrow 0$$

$$d + p \rightarrow {}^{3}He + \gamma$$

$$0 \rightarrow 0$$

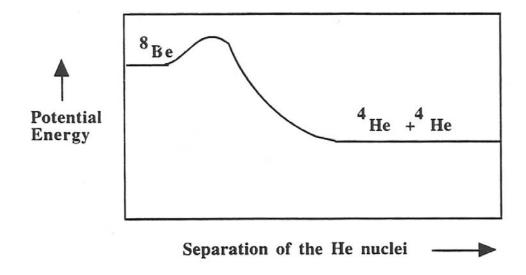
3
He + 3 He \rightarrow 4 He + 2 P

4
He = 2 p + 2 n

How to get past helium? We need C, O, N, P, S, ...

$${}^{4}\text{He} + {}^{4}\text{He} \rightarrow {}^{8}\text{Be} = 4p + 4n$$

Problem: ⁸Be has more nuclear potential energy than parts It is unstable (radioactive).



Need another ⁴He to hit ⁸Be before it falls apart

$${}^{4}\text{He} + {}^{4}\text{He} \rightarrow {}^{8}\text{Be}$$

$${}^{8}\text{Be} = 4p + 4n$$

8
Be = 4p + 4n

8
Be + 4 He \rightarrow 12 C + γ

$$\Rightarrow \Rightarrow \Rightarrow \Rightarrow \Rightarrow 12C = 6p + 6n$$

$${}^{2}C = 6p + 6n$$

$$^{4}\text{He} + ^{12}\text{C} \rightarrow ^{16}\text{O}$$

$$^{16}O = 8p + 8n$$

$$^{16}O + ^{16}O \rightarrow ^{32}S + \gamma$$

Sulfur

$$^{16}O + ^{16}O \rightarrow ^{31}P + p$$

Phosphorus

$$^{16}O + ^{16}O \rightarrow ^{28}Si + ^{4}He$$

Silicon

$${}^{4}\text{He} + {}^{4}\text{He} \rightarrow {}^{8}\text{Be}$$

$${}^{8}\text{Be} = 4p + 4n$$

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Be = 4p + 4n

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Be + 4 He \rightarrow 12 C + γ

$$\Rightarrow \Rightarrow \Rightarrow \Rightarrow \Rightarrow 12C = 6p + 6n$$

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$$^{4}\text{He} + ^{12}\text{C} \rightarrow ^{16}\text{O}$$
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$$^{16}O = 8p + 8n$$

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Sulfur

$$^{16}O + ^{16}O \rightarrow ^{31}P + p$$

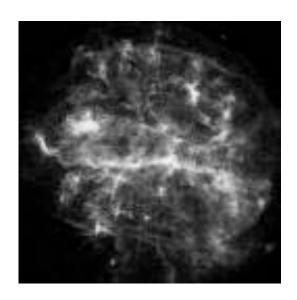
Phosphorus

$$^{16}O + ^{16}O \rightarrow ^{28}Si + ^{4}He$$

Silicon

Questions

- What was needed to make the bioelements?
- Are any missing?
- How do the bioelements get out of the star?



Summary

Heavy elements needed for life were created by

early generations of massive stars.

Except for H, we are made of star debris

Natural forces (Gravity, EM, Nuclear) produced first evolution of matter from simple to complex (p → heavy elements)