### Disappearance of Atoms

- As we go back in time, temperature goes up. - T=2.73(1+z) K
- At *z*~1100, *T*~3000 K
  - $-\,$  About the same temperature as M-dwarfs
- Ionization of hydrogen atoms
  - H + photon  $\rightarrow$  p + e<sup>-</sup>
    - Inverse process: recombination  $p + e^- \rightarrow H + photon$
- Electrons scatter photons in all directions
  - Similar to the interior of the Sun
  - Photons cannot travel freely
  - The universe becomes "opaque"
- Therefore,  $z \sim 1100$  is the edge of the "visible universe"
  - $-\,$  We cannot look back beyond this epoch using light
  - The epoch called the *decoupling epoch*.

## Radiation Era

- In the present universe, radiation energy/matter energy is about 1/3200.
- At the decoupling epoch, this ratio is about 1/3.
- At *z*~3200, the radiation energy equals the matter energy the matter-radiation equality
  - *z*>3200: Radiation Era
  - *z*<3200: Matter Era
- In the radiation era,
  - $\rho t^2 \sim 10^6$
  - $tT^2 \sim 10^{20}$ 
    - $\rho$  in g/cm<sup>3</sup>, *t* in seconds, and *T* in Kelvin.
- The radiation era ends at age about 30,000 years.
  - The decoupling epoch is  $\sim 380,000$  years.



## Nucleosynthesis

- The origin of helium
  - About 25% of the mass of atoms in the universe is helium.
    - 85% is hydrogen
  - Where did helium come from?
  - Stars fuse hydrogen into helium
    - This was incredibly inefficient process
    - Stars could fuse **only 2%** of all hydrogen in the universe into helium not enough!!
- The first three minutes
  - $T \sim 1$  billion K
  - The universe was a very efficient nuclear reactor
  - "Big-bang nucleosynthesis"

### Synthesize Helium

- $n + p \rightarrow D + photon$ 
  - When *T*>1 billion K, photo-dissociation process
    D + photon → n + p also occurs frequently and destroys the formed deuterium
  - One has to wait until temperature drops below 1 billion K: t~100 seconds
- $D + D \rightarrow T + p$ 
  - T is tritium (<sup>3</sup>H; one proton and two neutrons)
- $T + D \rightarrow {}^{4}He + n$

#### Deuterium

- The amount of deuterium depends on density of nuclei, or  $\Omega_{atoms}$ . How?
- Low  $\Omega_{\text{atoms}}$ 
  - Collision of deuterium,  $D + D \rightarrow T + p$ , does not occur frequently
  - A larger fraction of D survives

#### • High $\Omega_{atoms}$

– More collision  $\rightarrow$  A smaller fraction survives

# Consumption of Neutrons

- To form D, both n and p are required
  - However, n slowly decays via  $\beta\text{-decay:}$ 
    - $n \rightarrow p + e^{-} + v_e$
  - Therefore, n is outnumbered by p, and the amount of helium synthesized is essentially determined by the amount of neutrons
  - n:p = 1:7
- This n/p ratio explains He:H=1:3

•This is one of the most important predictions of the Big- after bang model synthesis

•The prediction is robust – it does not depend very much on cosmological parameters



#### ...yet we need stars...

- The Big-bang nucleosynthesis cannot synthesize any elements heavier than Lithium.
   – Density of nuclei was too low
- The initial proposal of the BBN was abandoned because of this "failure"
- Later it was realized that heavier elements must be synthesized by fusion in stars
  - Although temperature is low (10 million K vs 1 billion K), density is high (100 g/cm<sup>3</sup> vs 20 µg/cm<sup>3</sup>)