**Cosmic Evolution** 

## Part 1: Protons to heavy elements

Big Bang occurred 13.7 Billion yrs ago (13.7 x 10<sup>9</sup> yr)
Only fundamental particles existed for first few minutes

<u>Name</u>	<u>Symbol</u>	<u>Charge</u>	<u>Mass</u>
Proton	р	+	$1.7 \times 10^{-24} \text{ g}$
Neutron	n	0	$1.7 \times 10^{-24} \mathrm{g}$
Electron	е	_	$1 \times 10^{-27} \mathrm{g}$
Photon	γ	0	0
Neutrino	ν	0	~0(?)

Building blocks of nuclei but only one kind of nucleus Proton = nucleus of Hydrogen

### Energy of Motion (Kinetic Energy)

$$E = \frac{1}{2} \text{ mv}^2$$
 (if v not close to c)

Gas at Temperature T,

Avg. Energy 
$$E = \frac{3}{2} kT$$

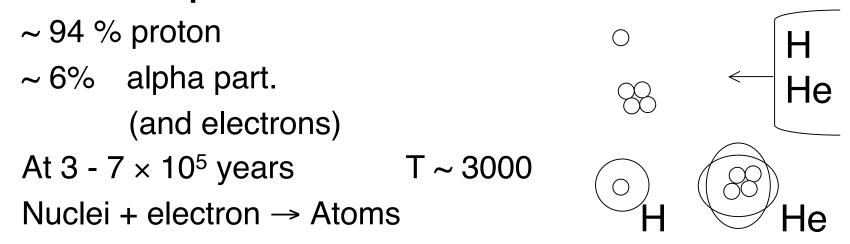
So avg. v : 
$$\frac{1}{2} \text{ mv}^2 = \frac{3}{2} \text{ kT}$$

$$V = \sqrt{\frac{3kT}{m}} = \left(\frac{3kT}{m}\right)^{\frac{1}{2}}$$

Higher  $T \rightarrow Higher v$ , E on avg.

Early Universe so hot that collisions broke apart any complex things that might have formed

As Universe expanded, T dropped at  $\sim 3$  min, T  $\sim 10^9$  K A few nuclei form (nucleosynthesis) at  $\sim 30$  min, T  $\sim 3 \times 10^8$  K end of nucleosynthesis Composition of Universe at 30 min.



#### First Generation Stars

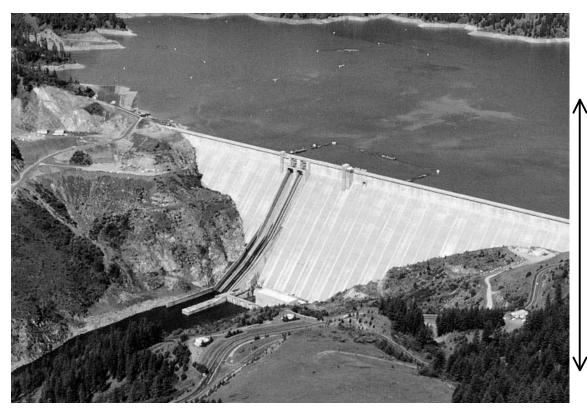
Expanding Universe
But, Gravity  $\rightarrow$  Galaxies ( $\sim 10^{11}$  to  $10^{12}$  M $_{\odot}$ )  $\rightarrow$  Stars (0.1 to 100 M $_{\odot}$ )

Oldest stars in disk ~ 10<sup>10</sup> y old

First generation stars → No C, O, N, ... ⇒ No life No Si, Fe ⇒ No Earthlike planets But they made "heavy" elements So later stars could have solid planets, life

## **Gravitational Potential Energy**

For example: Reservoir of water behind dam

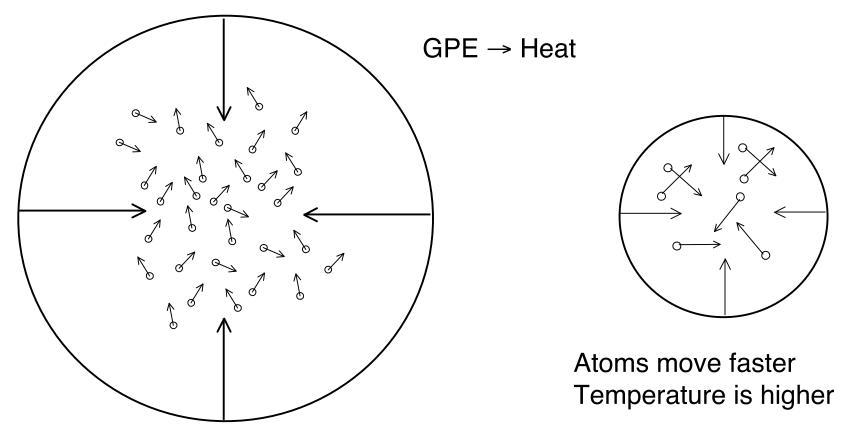


Energy released as water falls

Lower GPE

#### Consider a clump of Gas collapsing to form a star

Apply to collapsing gas



Temperature in core reaches 10<sup>7</sup> k

Nuclear reactions begin

Collapse stops

Why?

# **Nuclear Potential Energy**

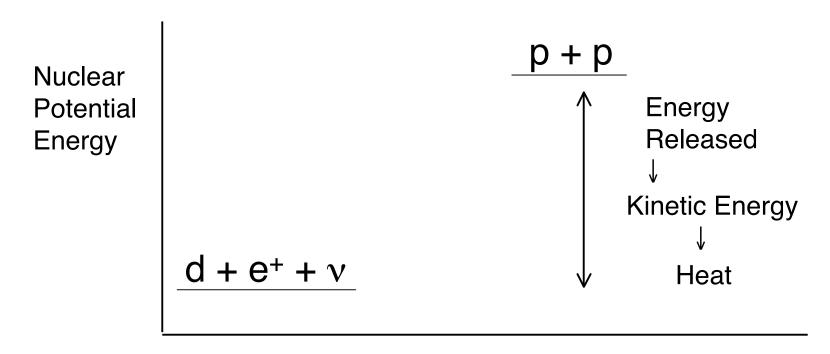
Four basic forces: gravity, electromagnetic, weak and strong nuclear force

Each has potential energy. Nuclear potential energy can be released by nuclear reactions.

e.g. 1st step: 
$$p + p \rightarrow d + e^+ + v$$

d = deuteron = proton + neutron e<sup>+</sup> = positron (antiparticle of electron)

## **Nuclear Potential Energy**



Separation of two protons

The energy released by nuclear reactions supplies heat → pressure

Resists gravity ⇒ stable star

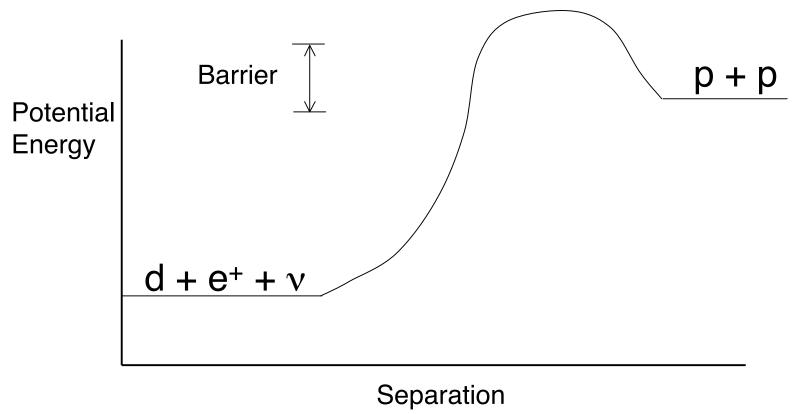
## **Electromagnetic Barrier**

Why do we need high T ( $\sim 10^7$  K)?

Protons have positive electric charge Like Charges Repel

As protons approach, repulsion grows corresponds to climbing hill of electromagnetic potential energy

## **Electromagnetic Barrier**



Barrier is really much higher than  $\frac{3}{2} k \cdot 10^7 k$ Very few can get over barrier

⇒ Stars live a long time rather than exploding



#### Questions

- Why do nuclear reactions produce a longlived system in star, but an explosion in a bomb?
- What will happen when a star's fuel runs out?

n

neutrino

$$p + p \rightarrow d + e^{+} + v$$

$$0 \rightarrow 0$$

$$d + p \rightarrow {}^{3}He + \gamma$$

$$0 \rightarrow 0$$

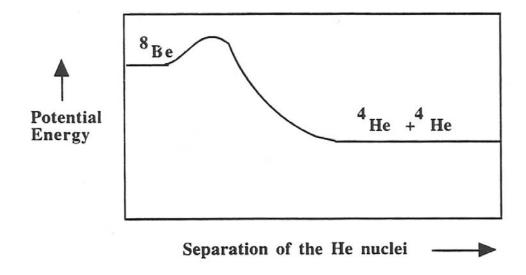
$$^{3}$$
He +  $^{3}$ He  $\rightarrow$   $^{4}$ He +  $^{2}$ P

$$^{4}$$
He =  $^{2}$ p +  $^{2}$ n

How to get past helium? We need C, O, N, P, S, ...

$${}^{4}\text{He} + {}^{4}\text{He} \rightarrow {}^{8}\text{Be} = 4p + 4n$$

Problem: <sup>8</sup>Be has more nuclear potential energy than parts It is unstable (radioactive).



Need another <sup>4</sup>He to hit <sup>8</sup>Be before it falls apart

$${}^{4}\text{He} + {}^{4}\text{He} \rightarrow {}^{8}\text{Be}$$

$${}^{8}\text{Be} = 4p + 4n$$

$$^{8}$$
Be = 4p + 4n

$$^{8}$$
Be +  $^{4}$ He  $\rightarrow$   $^{12}$ C +  $\gamma$ 

$$\Rightarrow \Rightarrow \Rightarrow \Rightarrow \Rightarrow 12C = 6p + 6n$$

$${}^{2}C = 6p + 6n$$

$$^{4}\text{He} + ^{12}\text{C} \rightarrow ^{16}\text{O}$$

$$^{16}O = 8p + 8n$$

$$^{16}O + ^{16}O \rightarrow ^{32}S + \gamma$$

Sulfur

$$^{16}O + ^{16}O \rightarrow ^{31}P + p$$

**Phosphorus** 

$$^{16}O + ^{16}O \rightarrow ^{28}Si + ^{4}He$$

Silicon

$${}^{4}\text{He} + {}^{4}\text{He} \rightarrow {}^{8}\text{Be}$$

$${}^{8}\text{Be} = 4p + 4n$$

$$^{8}$$
Be = 4p + 4n

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$$^{16}O + ^{16}O \rightarrow ^{31}P + p$$

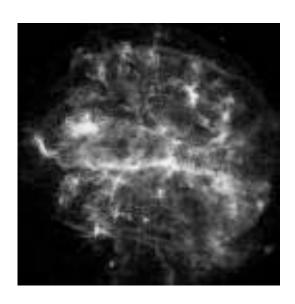
Phosphorus

$$^{16}O + ^{16}O \rightarrow ^{28}Si + ^{4}He$$

Silicon

### Questions

- What was needed to make the bioelements?
- Are any missing?
- How do the bioelements get out of the star?



## **Summary**

Heavy elements needed for life were created by

early generations of massive stars.

Except for H, we are made of star debris

Natural forces (Gravity, EM, Nuclear) produced first evolution of matter from simple to complex (p → heavy elements)