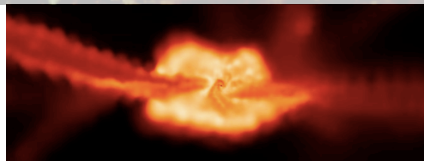


Molecules and dust in the early universe: the supernova connection

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Overview

- Evidence for dust in supernovae
- Chemical kinetics versus other dust synthesis formalisms
- What kind of dust ? Chemical routes
- Models for primitive supernovae
- Molecules and dust in PISNe
- Depletion efficiencies and mixing
- Conclusions

Evidence for molecules and dust in Supernovae

Dust in nearby Type II supernovae:

- SN1987A : IR detection of **CO, SiO** and dust (from ~ **350 days** to ~ **800 days** post-explosion) - $10^{-4} - 10^{-3} M_{\text{sun}}$ (Roche et al. 1991, Meikle et al. 1993, Ercolano et al. 2007)
- SN2003gd: dust observed with Spitzer - $10^{-2} M_{\text{sun}}$ (Sugerman et al. 2006)
- No large amounts of dust detected in SN2004dj - $4 \times 10^{-5} M_{\text{sun}}$ (Meikle et al. 2007), or SN2004et ~ $10^{-4} M_{\text{sun}}$ (Kotak et al. 2009) – detection of **CO and SiO**.
- Dust amounts observed in SNRs: Cas A ~ $0.1 M_{\text{sun}}$ in IR+Submm
- From isotopic anomaly analysis of meteoritic grains: stardust inclusions formed in Type II SN ejecta – graphite, silicates, SiC, and Si_3N_4 (Zinner 2006)

-Are local CCSNe efficient dust makers? $1e^{-1}-1e^{-6} M_{\text{sun}}$
-Silicate, metal oxide and/or carbon grains or a mixture?

At high redshift (< 1 Gyr or $z > 6$)

- Large amount of dust deduced from reddening of background quasars:
J1148+5251 at $z = 6.4$: $M_{\text{dust}} = (0.9 - 4) \times 10^8 M_{\text{sun}}$ - $M_{\text{dust}}/M_{\text{gas}} = 0.5 - 1 \times 10^{-2}$
- Dust mass implies a **dust yield of ~ $1 M_{\text{sun}}/\text{SN}$** (Dwek et al. 2007).
- Difficult to reconcile with local CCSNe.

Chemical kinetics versus other formalisms

- **Thermodynamic equilibrium** used in meteorite studies (Fedkin et al. 2010):
P-T phase diagram inappropriate for dynamical outflows out of equilibrium.
- **Classical nucleation theory**: dust formation in SNe and AGB stars
(Draine 1979, Kozasa et al. 1989, Todini & Ferrara 2001, Nozawa et al. 2003, Schneider et al. 2004, Ferrarotti & Gail 2006)

involves concepts like **surface tension, sticking coefficient, supersaturation ratio, equilibrium distribution of critical clusters...**
can't apply to Å size dust precursors out of equilibrium (Donn & Nuth 1985)

... Ignore the chemical synthesis of molecules and dust precursors from the gas phase...

Chemical kinetic approach:

Nucleation of gas phase dust precursors + condensation (coagulation+surface growth)

Couple the gas phase to the solid phase

Very **general** and **powerful** approach which can be applied to any stellar outflow...

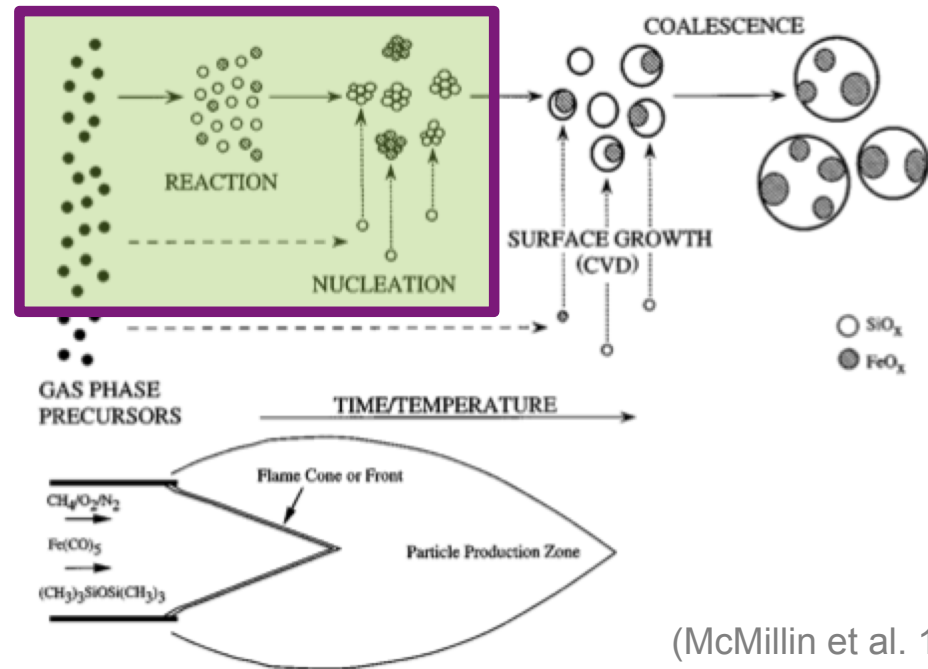
Chemical kinetics versus other formalisms

In the laboratory, we can form dust from the gas phase using different techniques:

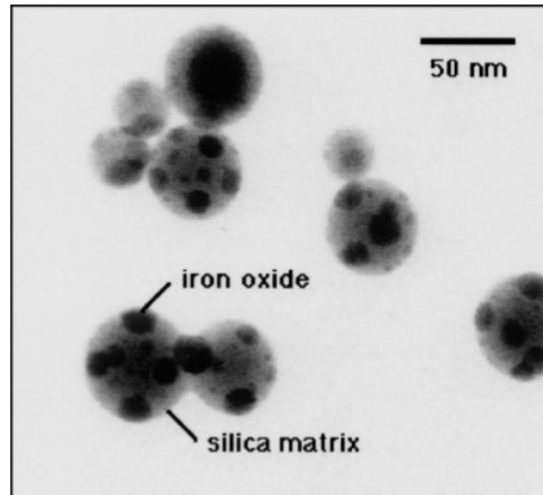
- Vaporization of solid rods
- Pyrolysis of hydrocarbons
- Flame aerosol reactors

Study the synthesis of **soot**, **metal oxides**, **silicate**, **metal carbide dust**...

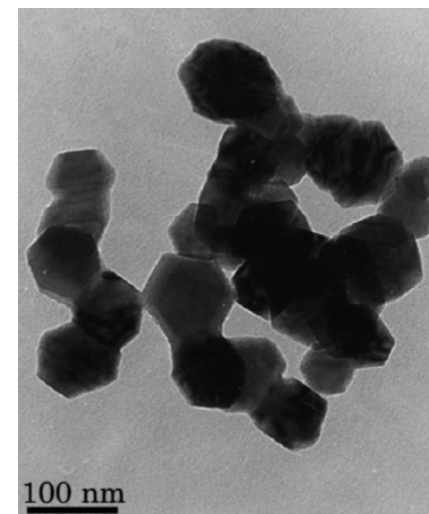
(Kaito et al. 2003, Jäger et al. 2009)



Fe_2SiO_4 - Fayalite



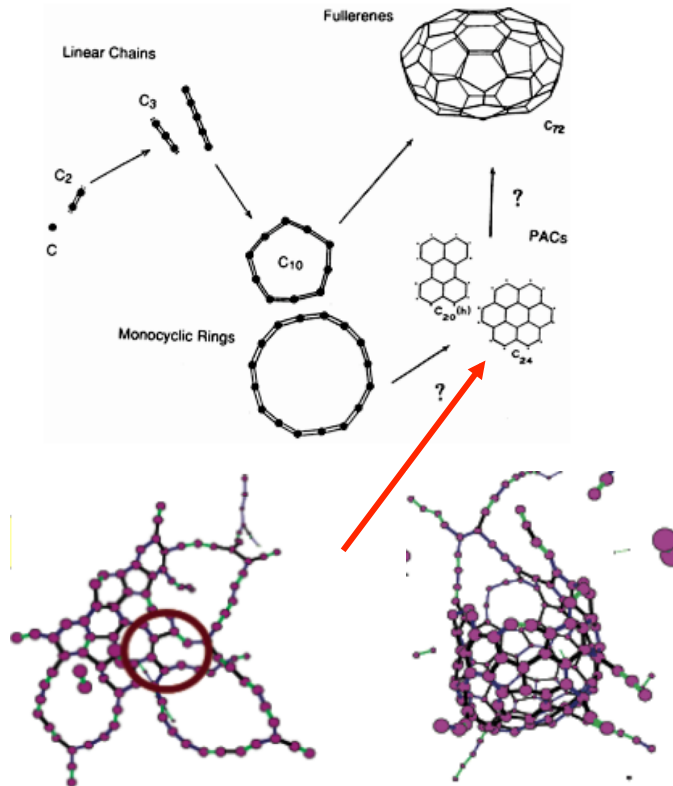
FeS - Troilite



What kind of dust? Some chemical routes

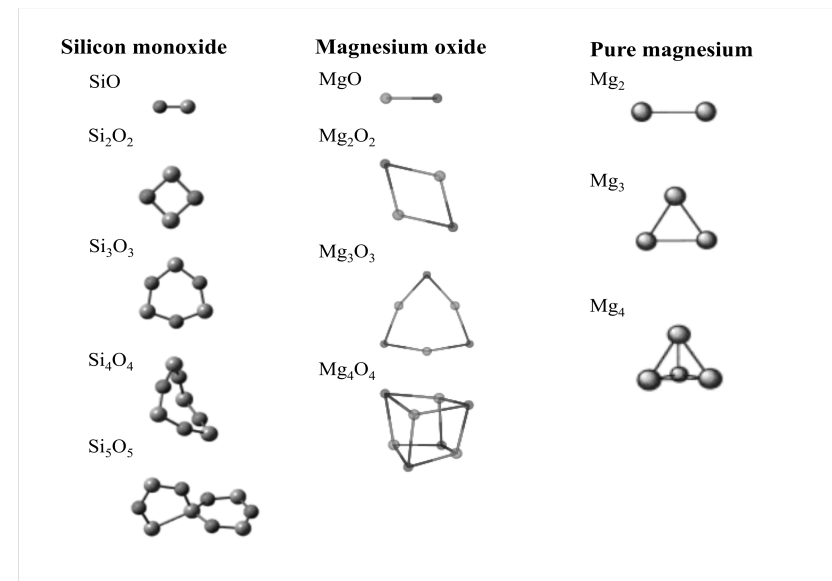
Amorphous carbon:

Without hydrogen: carbon chains & rings



(Cherchneff et al. 2000, Irle et al. 2003)

Silica, metals and oxides

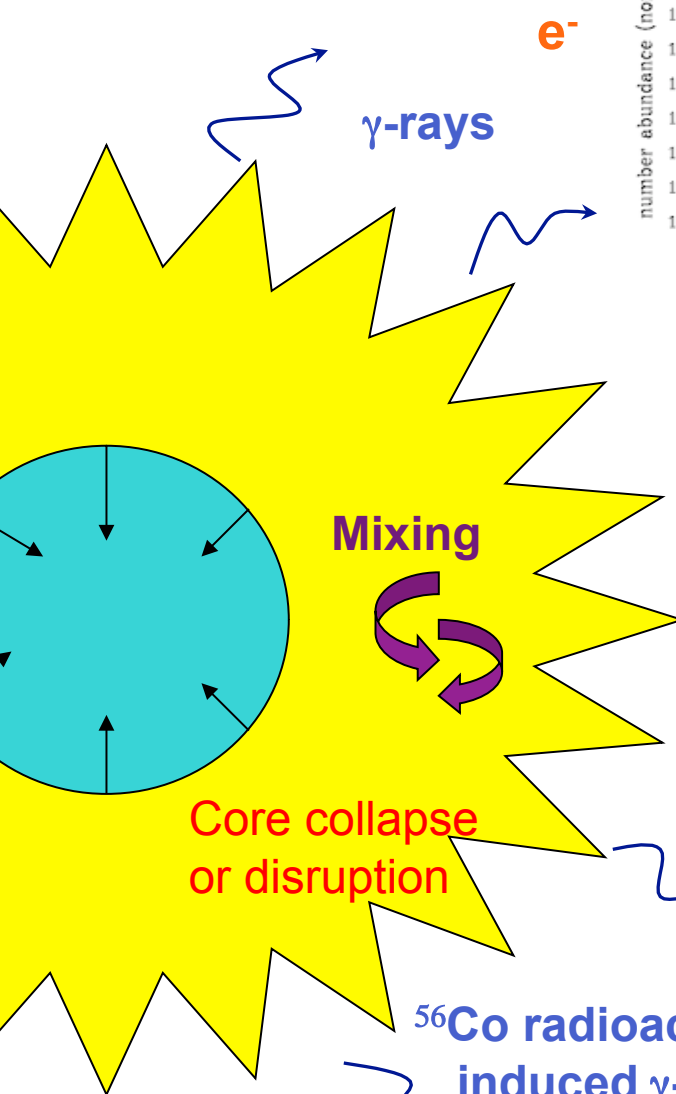
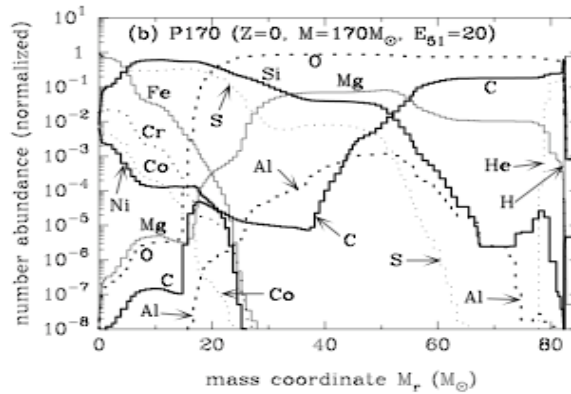


- Slow radiative association reactions for **carbon chains** formation from **free C** (Clayton et al. 1999, 2001)
- Pressure dependent nucleation rate for silica (Zachariah & Tsang 1993)
- Not all chemical steps and intermediates are characterized → rely on experiments (kinetics & intermediate identification) and theory (structure, TS)

No a priori assumption on what kind of dust forms

All chemical routes apply to AGBs, Supergiants, Wolf-Rayet CWR, R CrB stars, or SN ejecta

20 - 170 - 270 M_{sun}



$\sim 2000-5000 \text{ kms}^{-1}$

CO, SiO

Dust - large amount (?)

H diffusion

H-rich progenitor envelope

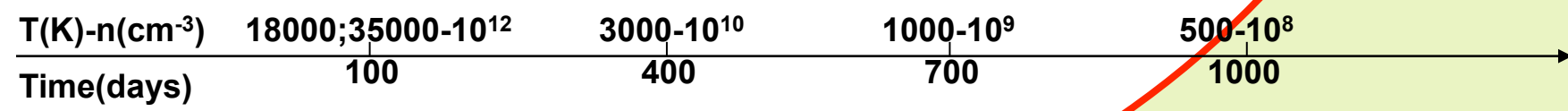
$n(e^-)/n_{\text{gas}} \leq 10^{-2}$

Blast wave

fast Compton electrons

UV

^{56}Co radioactivity-induced γ -rays



Primitive PISN and CCSN models

Chemical kinetic network: ~ 80 species and ~ 500 reactions with measured/theoretical rates from **astrochemistry**, **aerosol** and **combustion chemistry**, and **material science**.

Formation processes

- Tri- and bi-molecular reactions (neutral-neutral and ion-molecules),
- Radiative associations (e.g., carbon chains)

Destruction processes

- Radioactivity-induced Compton e^- destruction (dissociation, ionization),
- Thermal fragmentation (high T),
- UV photodissociation and ionization,
- Dissociative recombination of ions and ion attack (O^+ , He^+)

Species **Linear, chains, rings** from H, O, C, N, Si, S, He, Al, Mg, and Fe.

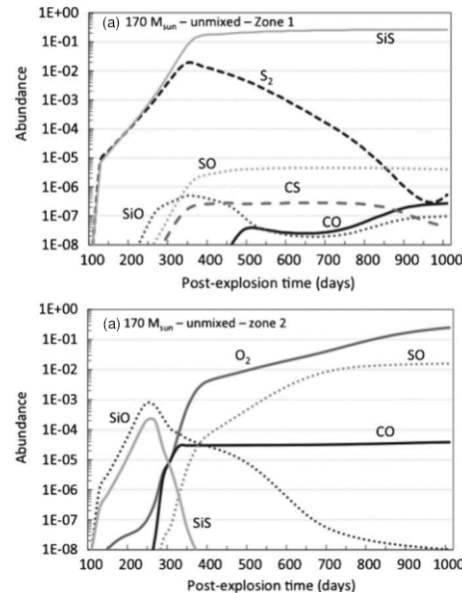
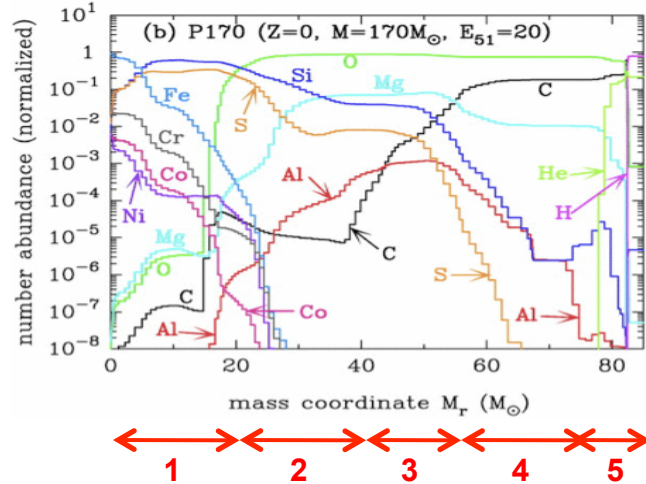
- CO, SiO, SO, CO₂, SiS, SiO₂, O₂, ions (CO⁺, SiO⁺, and all metals)
- (FeO)_n, (MgO)_n, (SiO)_n, (SiO₂)_n, AlO, (Mg)_n, (Fe)_n, (Si)_n, (FeS)_n, (MgS)_n; n=1-4
- Carbon chains [C₂ – C₉] and ring C₁₀

Two ejecta mixing extreme cases:

- **fully microscopically mixed He core** (Umeda & Nomoto 2002, Heger & Woosley 2002),
 - **stratified He core where each zone is microscopically mixed** (Nozawa et al. (2003),
- **Temperature and density** profiles from Nozawa et al. (2003).
- **Velocity** : $v = 2000\text{-}5500 \text{ km s}^{-1}$.

Results: molecules and dust clusters in PISNe

1 - 170 M_{sun} unmixed



- Chemistry **is not** at **steady-state**
- A 170 M_{sun} ejecta forms **$\sim 35 M_{\text{sun}}$** of molecules at day 1000:

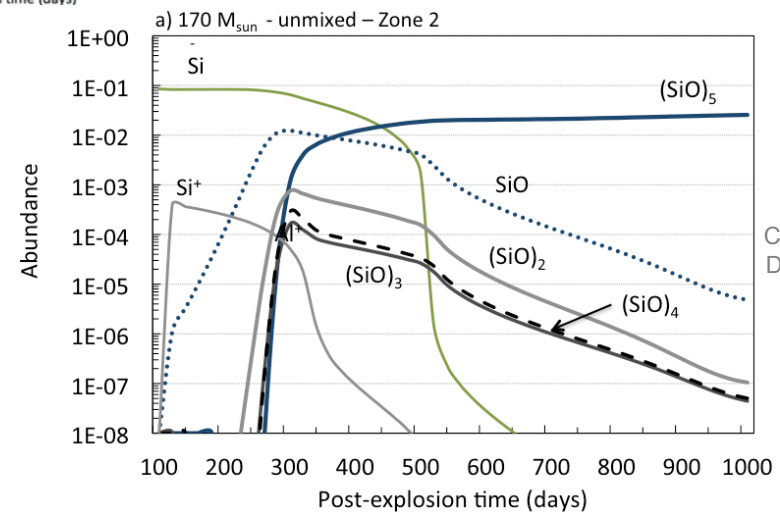
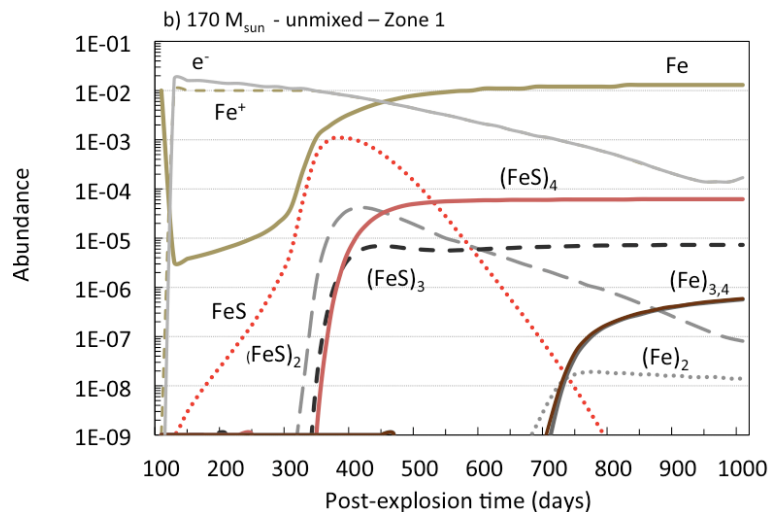
O_2 ($15M_{\text{sun}}$),

CO ($11M_{\text{sun}}$),

SiS ($8M_{\text{sun}}$),

SO ($1M_{\text{sun}}$).

(Cherchneff & Lilly 2008, Cherchneff & Dwek 2009)

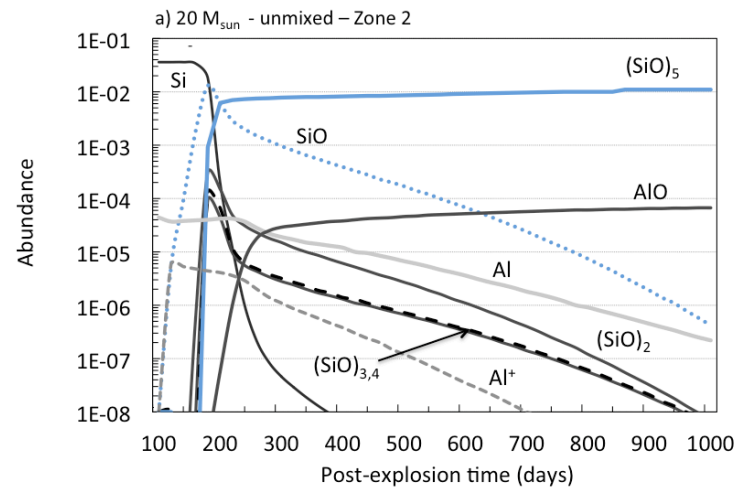
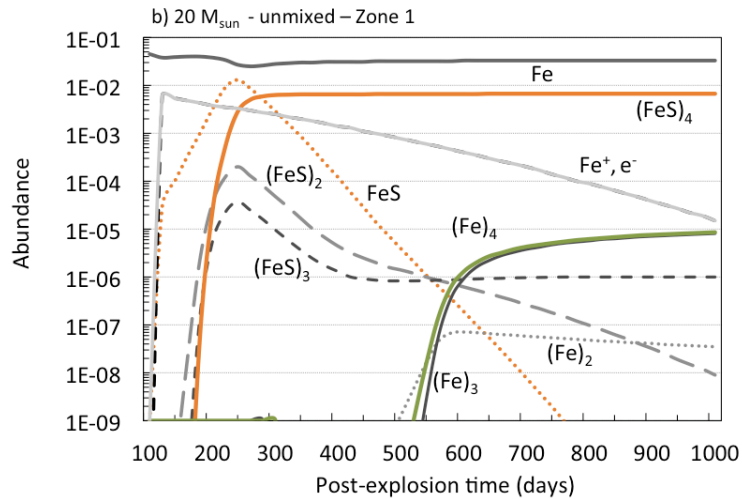


Cherchneff & Dwek (2010)

- **$\sim 6 M_{\text{sun}}$** of Silica (SiO_2)/silicate (forsterite Mg_2SiO_4), pure Si dust (99.5%), troilite (FeS) and alumina (Al_2O_3) (0.5%) is ejected at 1000 days equivalent to **3.5 %** of progenitor mass,
- **no carbon dust** forms in Zones 4 & 5 due to **oxidation of chains by O** and **dissociation by He^+ attack**.

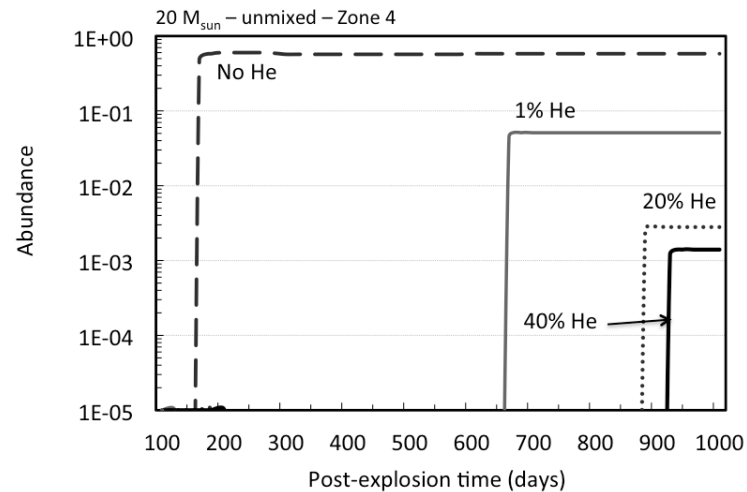
Results: molecules and dust clusters in PISNe

1 - 20 M_{sun} unmixed



Cherchneff & Dwek (2010)

Zone 4: He-rich; C/O ~ 26

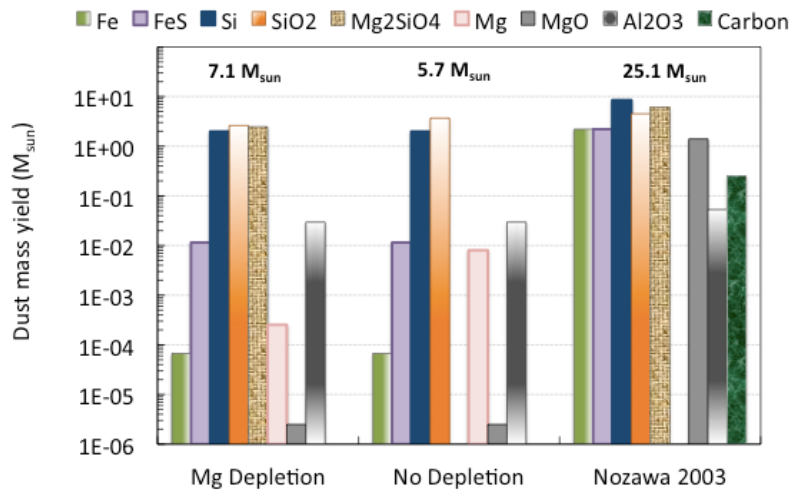


C_{10} abundance

Carbon ring C_{10} forms in Zones 4 (C/O > 1) if He^+ absent ($\sim 0.014 M_{\text{sun}}$)

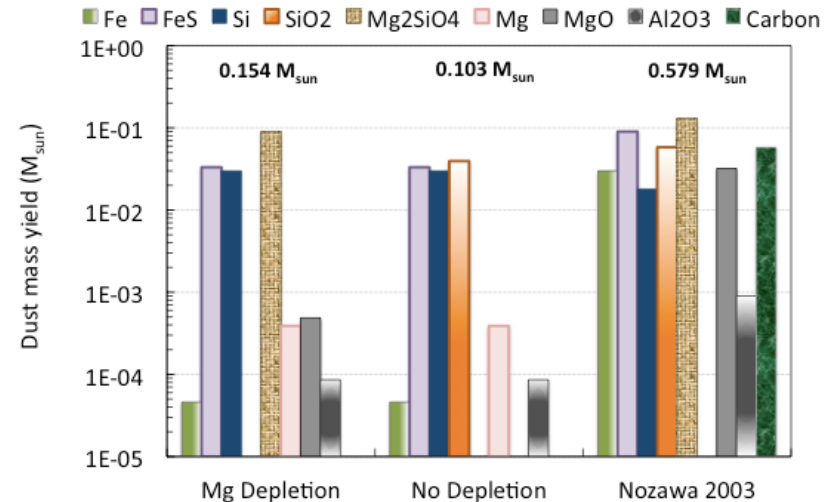
Results: molecules and dust precursors

170 M_{sun} progenitor unmixed ejecta



(Cherchneff & Dwek 2010)

20 M_{sun} progenitor unmixed ejecta



	Molecules (M_{sun})	Dust (M_{sun})	
		Depletion	No depletion
Unmixed 170 M_{sun}	35	7.1	5.7
% of prog. Mass	20.6%	4.2%	3.3%
Unmixed 20 M_{sun}	1.5	0.154	0.103
% of prog. Mass	7.5%	0.8%	0.5%

Existing models:

Nozawa et al. 2003: for a PISN, $M_{\text{dust}}/M_{\text{prog}} \sim 15 - 30\%$ - for 20 M_{sun} CCSN, $M_{\text{dust}}/M_{\text{prog}} \sim 3 - 4\%$

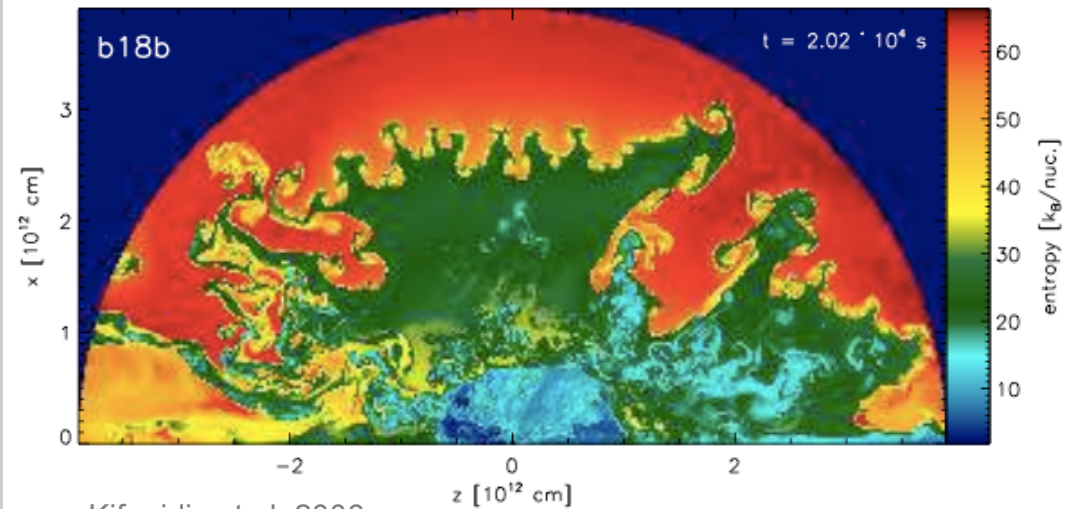
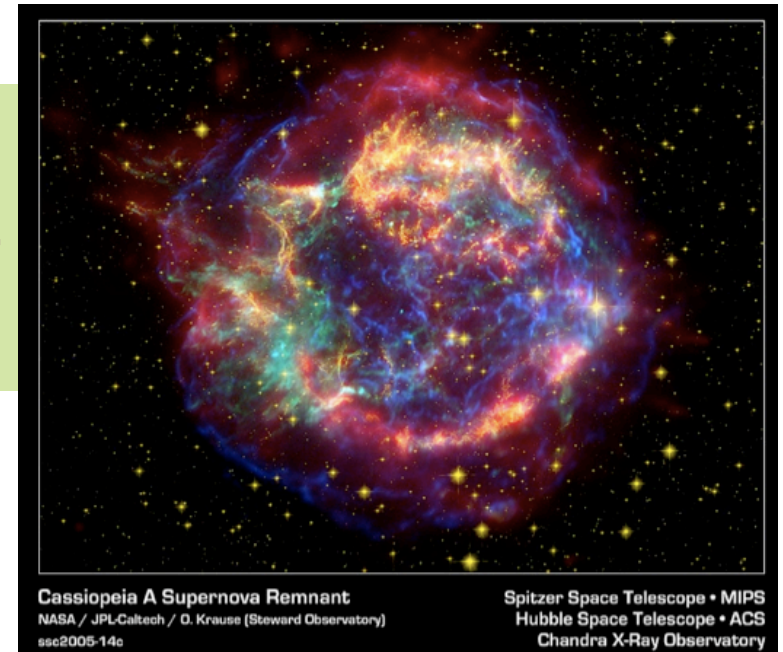
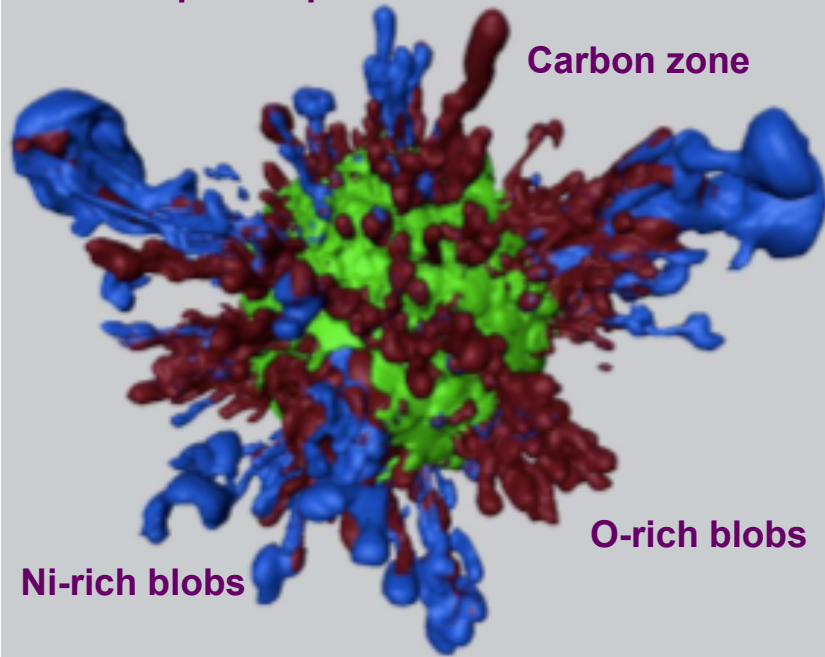
- Our dust upper limits are **smaller by a factor ~5** than dust budgets from existing models
- Dust clusters imply a **different dust chemical composition**
- Chemistry and molecular formation act as a **bottleneck** to dust formation

Depletion efficiencies and mixing

- Observational/theoretical evidence for mixing in local SNe and SN remnants
- Evidence for microscopic mixing of non-adjacent layers from SiC and Si₃N₄ inclusions in meteorites

(Hammer et al. 2009)

2.5 hour post explosion



Kifonidis et al. 2006

Depletion efficiencies and mixing

Depletion Efficiencies of Metals in Molecular Clusters and Molecules (Underlined) for the 170 and 20 M_{\odot} Unmixed Ejecta^a

Element	Clusters	170 M_{\odot}					20 M_{\odot}			
		Zone 1	Zone 2	Zone 3	Zone 4	Zone 5	Zone 1	Zone 2	Zone 3	Zone 4
Si	(SiO ₂) ₅	...	<u>50.0%</u>	<u>50.0%</u>	<u>46.7%</u>	37.2%	0.12%	49.5%	48.2%	...
	(Si) _{4,5}	1.8%	<u>50.0%</u>	<u>50.0%</u>	<u>46.7%</u>	37.2%	3.1%	49.5%	48.2%	...
	SiS	<u>26.5%</u>	<u>28.3%</u>
O	(SiO ₂) ₅	...	9.6%	2.7%	3.9%
	O ₂	...	38.0%	46.4%	12.3%	60.9%	27.6%	...
	CO	29.1%	43.0%	32.7%	...
	SO	...	1.2%	0.3%	0.1%	...	0.7%
S	(FeS) ₄	0.09%	8.7%
	SiS	98.9%	91.4%
	SO	...	100%	99.8%	99.2%
Al	AlO	...	97.0%	98.3%	92.1%	99.1%	96.8%	...
Fe	(FeS) ₄	1.8%	44.0%
	(Fe) ₄	0.02%	0.1%
Mg	(Mg) ₄	...	<u>0.1%</u>	<u>2.3%</u>	<u>0.1%</u>	0.91%	0.04%	...
	(MgO) ₄	0.73%	...
C	C ₁₀	0%–21.6% ^b	0%–95.3% ^b
	CO	100.0%	100.0%	77.3%	...	99.8%	99.7%	0%–0.14% ^b

- Si is **totally depleted** by silica/silicate dust – **SiO line decline observed in local SNe**
- **Pure Fe dust does not form efficiently** even in the absence of sulphur. In the presence of sulphur, Fe depletion **in FeS for CCSNe**.
- Carbon trapped in CO in PISNe but can be **depleted in AC dust for CCSNe** if He⁺ is absent.
- **Chemical depletion efficiencies and mixing provide a natural limitation to dust formation.**

Conclusions

- **Molecules** are abundant in primordial **massive** supernova ejecta with $\sim 35 M_{\text{sun}}$ released to the local gas,
- Our primordial SNe form **less** dust (**1/5**) than in existing models because of the **'chemical bottleneck'**,
- For PISNe, essentially **silica/silicates and pure Si grains. No carbon.**
—→ other carbon dust providers at $z > 6$? BELCs in quasar winds. WC colliding winds. AGB stars...,
- **Carbon dust** can only form in primitive CCSNe if the environment is **He⁺ free** – realistic?
- Mixing and chemical depletion efficiencies are **natural limiting agents** to dust formation and mass yields,
- But PISNe are definitely **molecule and dust providers** to the early universe,

Open questions: Molecule & dust survival in clumps? Sources of **cooling** for PopIII.2 & Pop II.5 star formation? Extinction properties of first galaxies?