

# Habitable Planets: Part 1

## Estimating $n_p$

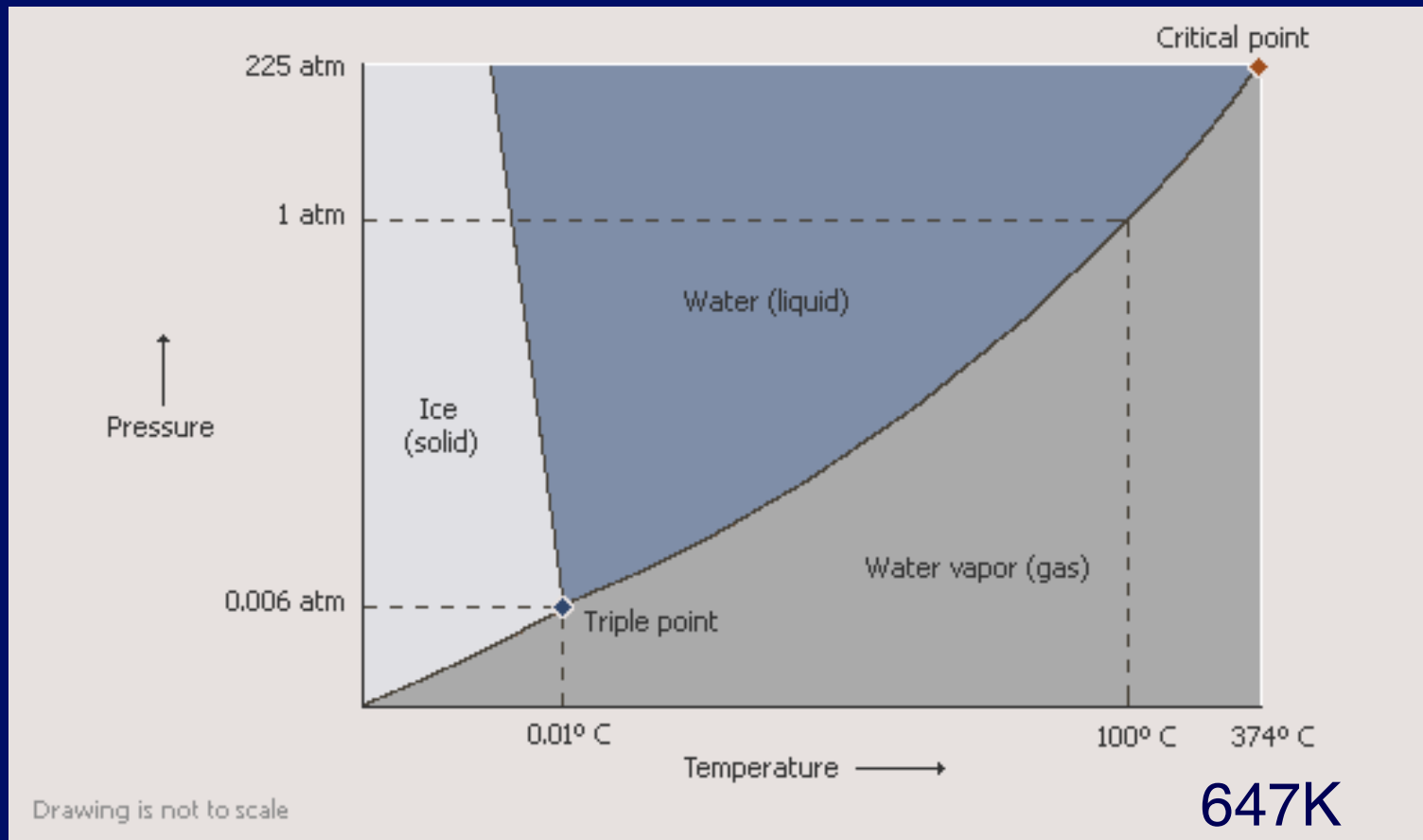
# General Considerations

- Number of planets, per planetary system, suitable for life ( $n_e$ )
- Useful to break into 2 factors
  - $n_e = n_p \times f_s$
- $n_p = n_e$  for stars like Sun
- $f_s$  is fraction of stars that are suitable
- $f_s \leq 1$ , but  $n_e$ ,  $n_p$  can be  $> 1$

# Key Requirement: a Liquid

- Need a liquid for a solvent
- Liquid phase needs particular range of temperature, pressure
  - Pressure depends on gravity, hence mass
  - Water ( $\text{H}_2\text{O}$ ) 273-373 K at Earth pressure
    - Up to 647 K at higher pressure
  - Methane ( $\text{CH}_4$ ) 91-109 K at Earth pressure
  - Ammonia ( $\text{NH}_3$ ) 195-240 K at Earth Pressure

# Water Phase Diagram



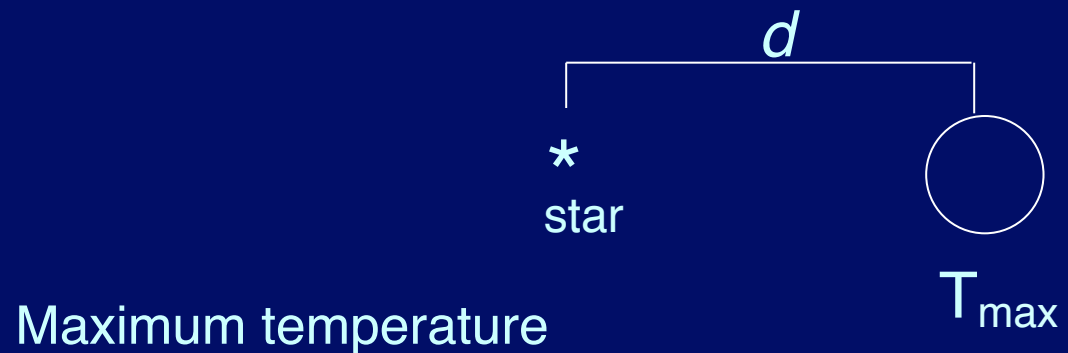
# What Sets the Temperature?

- For solid objects in space
  - Absorption and emission of light
  - Heats and cools the object
  - For a “blackbody”
    - Absorption  $\propto L/d^2$
    - Emission  $\propto T^4$
    - Radiative Balance
      - Energy absorbed = Energy emitted



# Planet Temperatures

1st approximation: A blackbody at a distance  $d$  from a star of luminosity  $L$



$$T = 394 \text{ K } (L / d^2)^{1/4}$$

$$T \propto L^{1/4} d^{-0.5}$$

$L$  in  $L_{\odot}$   
 $d$  in AU

# Question

Question: Planet 1 has a temperature of 300 K. There is an identical second planet 4 times as far from the star as planet 1. What will be the temperature of planet 2?

2nd approximation: A fraction of the light is reflected (not absorbed)

Call this fraction the albedo (A)

$$T = 394 \text{ K} \left[ \frac{(1-A) L}{d^2} \right]^{1/4}$$

e.g. Moon  $A = 0.07$   $T_{\max} = 387$   
 $L = 1 L_{\odot}$  correct to few %

But Earth :  $A = 0.39 \Rightarrow T_{\max} = 342 \text{ K}$  predicted  
 $T_{\max} \leq 313 \text{ K}$



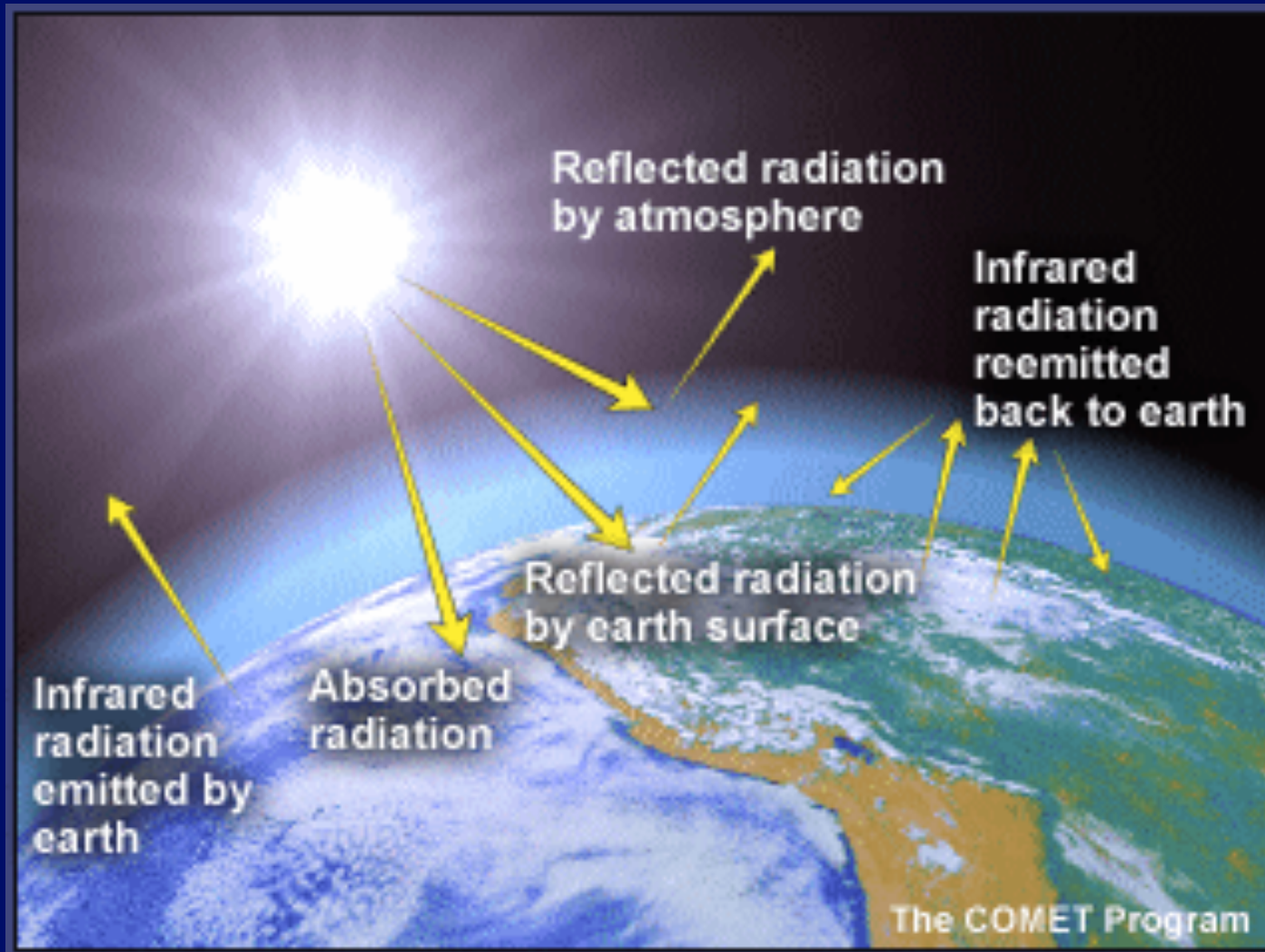
3rd approximation:

Account for rapid rotation -  $T_{\max}$  less  
 $T_{\min}$  more  
close to  $T_{\text{avg}}$

$$T_{\text{avg}} = 279 \text{ K} \left[ \frac{(1-A) L}{d^2} \right]^{1/4}$$

Earth:  $A = 0.39 \Rightarrow$   $T_{\text{avg}} = 246\text{K}$   
Actual  $T_{\text{avg}} = 288\text{K}$

# 4th approximation: Greenhouse effect



## Consequences of Greenhouse Effect:

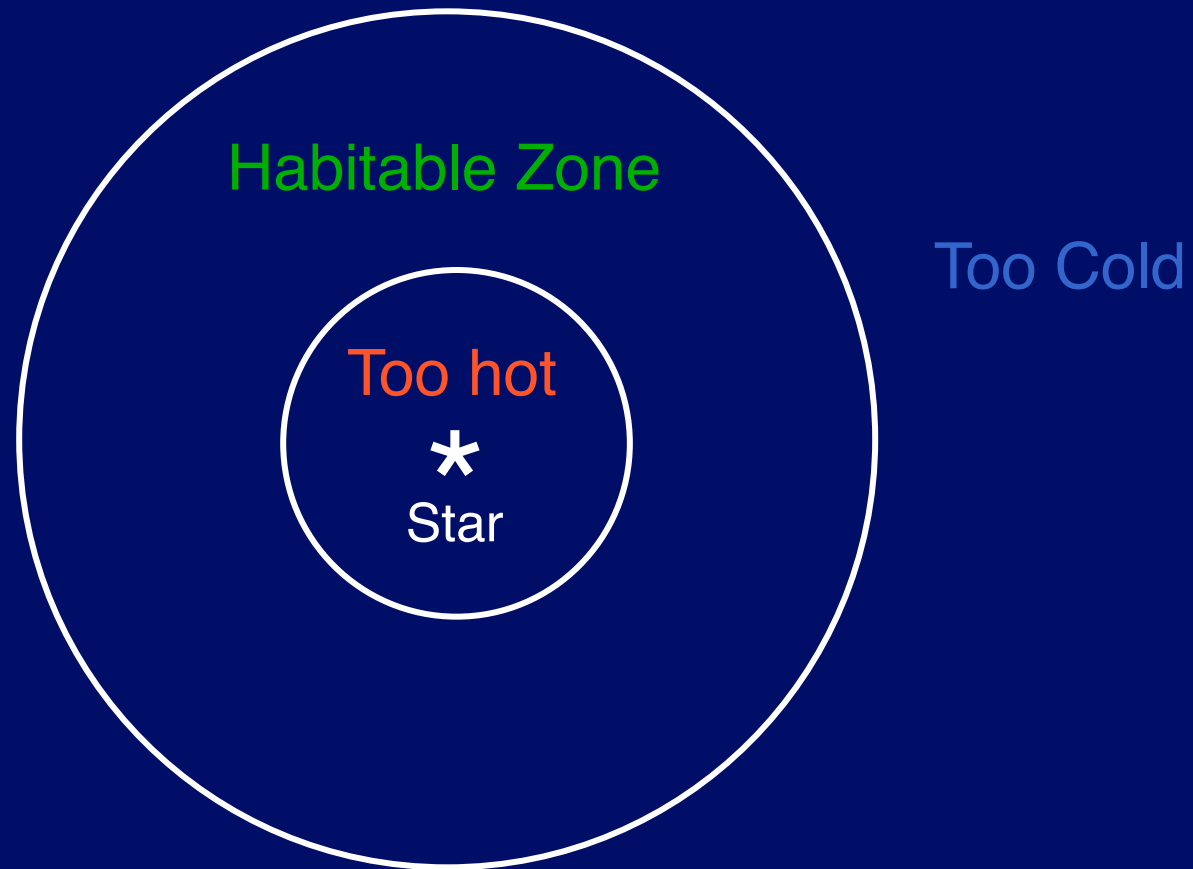
Raises  $T_{\text{avg}}$  (Earth) by about 40K

Otherwise  $T_{\text{avg}} < T_{\text{freeze}}$

⇒ Frozen Planet

# The Habitable Zone

- For fixed stellar luminosity, greenhouse effect
- A required temperature range
  - For example, liquid water
- Translates to required range of distances from the star



But Greenhouse Effect could have a big impact on the size and location of HZ

# Continuously Habitable Zone

- Nearly  $5 \times 10^9$  yrs for intelligent life on Earth
- CHZ is habitable for  $5 \times 10^9$  yr
- Stars increase  $L$  during main sequence
  - HZ moves out, CHZ is smaller than HZ at any given time
  - For example, current Earth would have been frozen over when Sun was young
  - Greenhouse effect must have been larger then

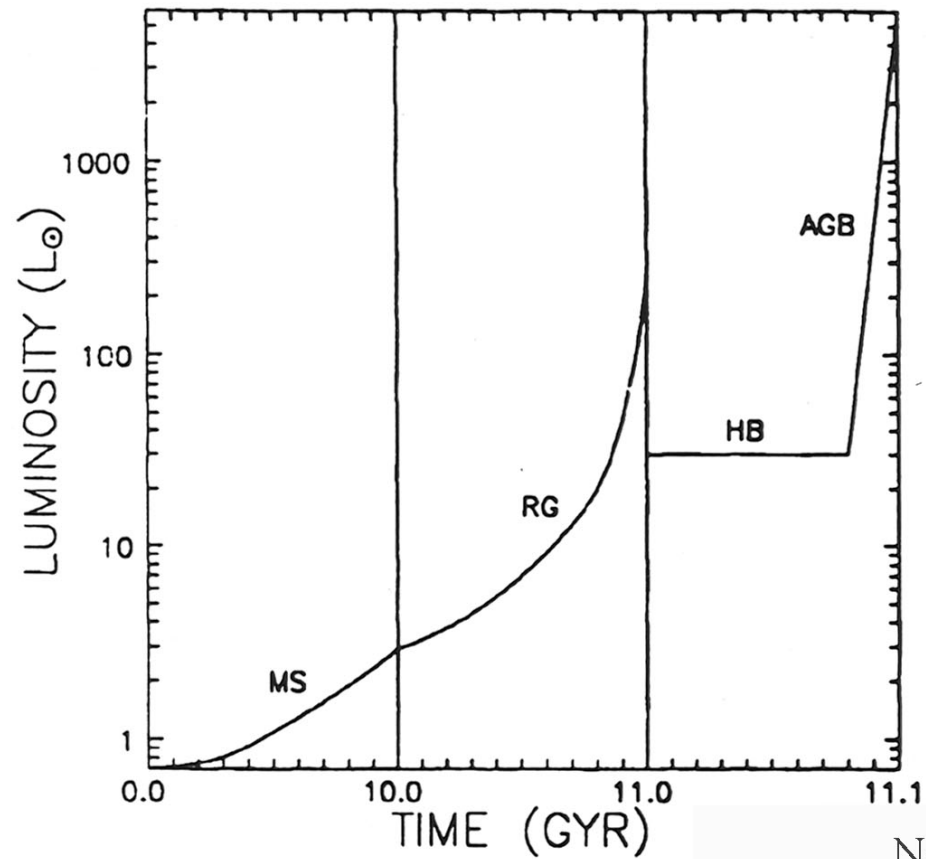


FIG. 1. Luminosity evolution of a  $1 M_{\odot}$  star of solar composition. Note that the scale is broken at 10 and 11 Gyr. The discontinuity at 11 Gyr is real and is due to the helium flash. The phases are: MS = main sequence, RG = red giant, HB = horizontal branch, and AGB = asymptotic giant branch. The final white dwarf phase is not illustrated. Details of the evolution during the HB and AGB phases are omitted.

$$r_2 = \sqrt{\frac{L(t)}{S_2(T)}}$$

Pessimist

## Computer Models

Hart

CHZ

0.95 - 1.01 AU

$$\Rightarrow n_p \lesssim 0.1$$

Middle of the Road

Negative feedback

→ thermostat

T

Rainfall

rock weathering



T ↓



CO<sub>2</sub>

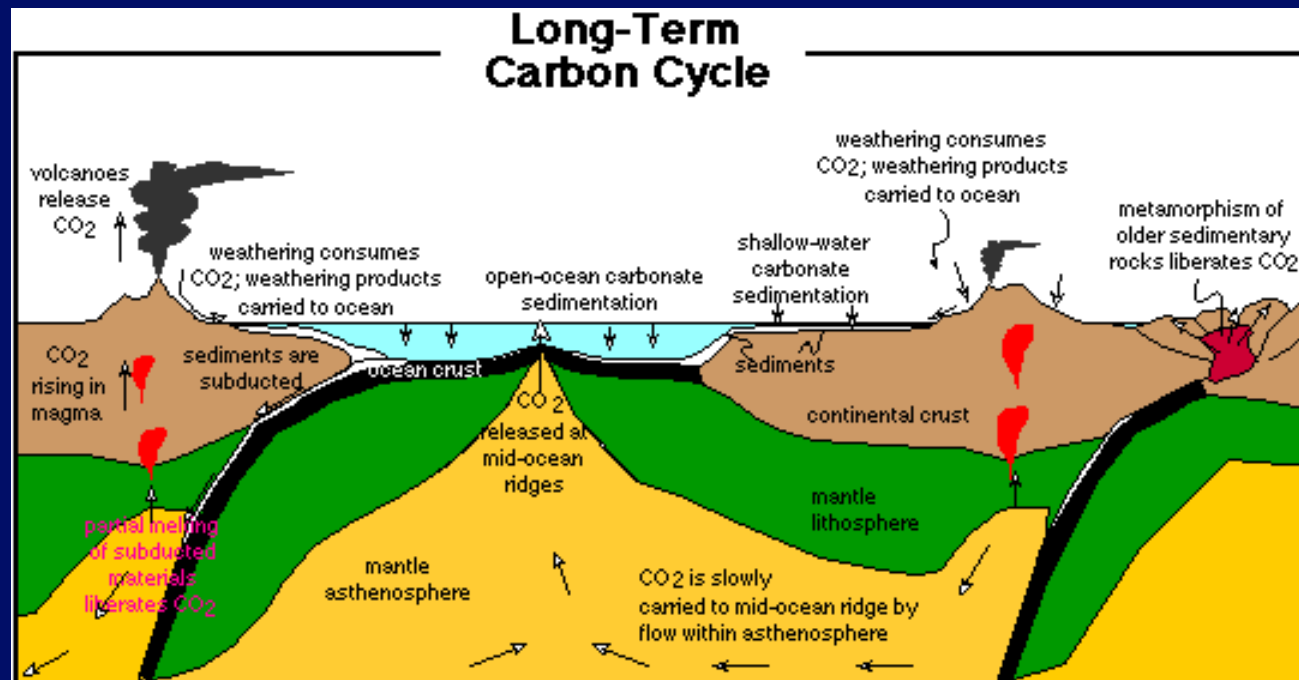
Whitmire et al.

CHZ 0.95 - 1.5 AU

$$\Rightarrow n_p \sim 1$$

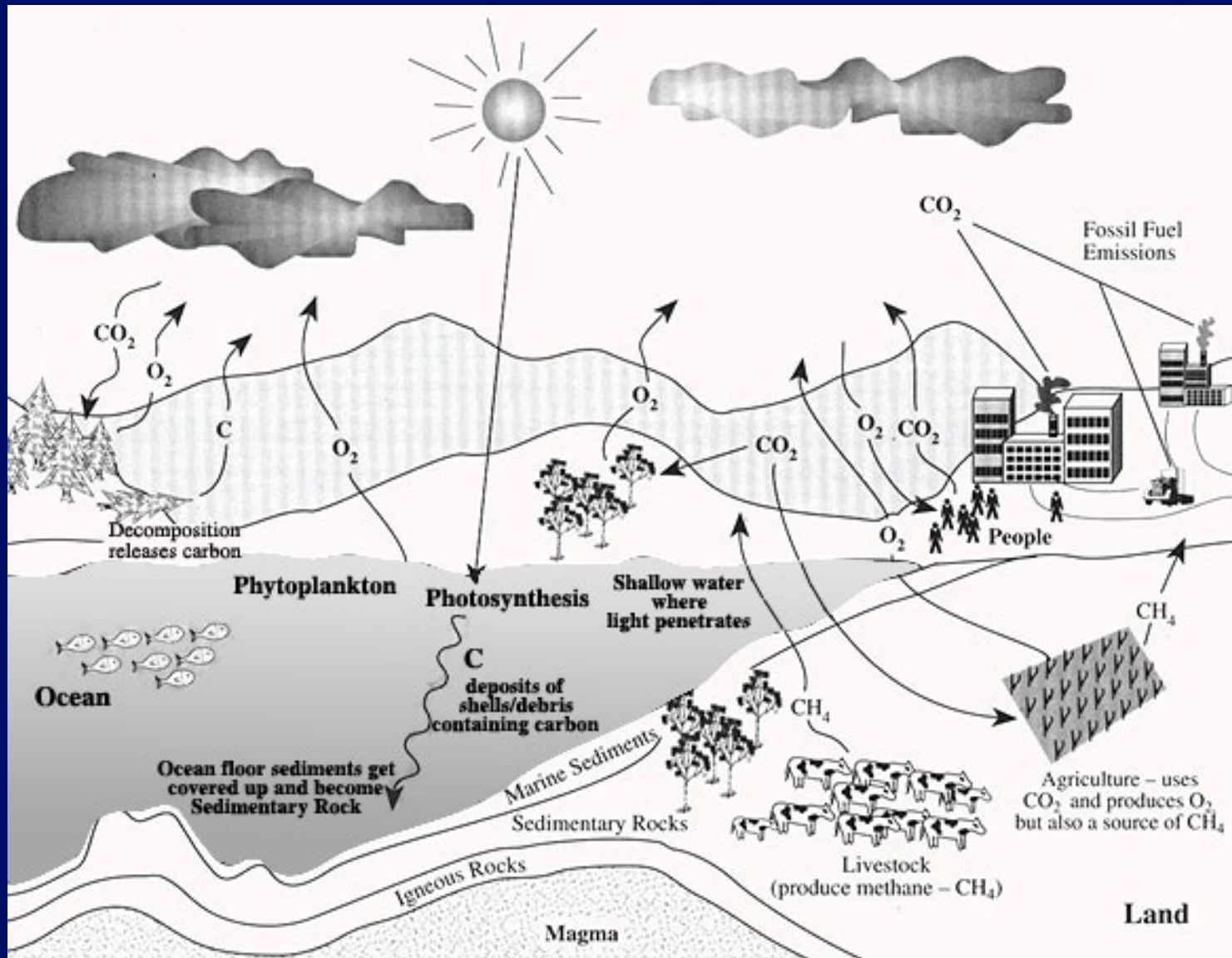


# The Carbon Cycle without Life



**Figure 7.3.** Schematic representation of the long-term global carbon cycle showing the flows (hollow arrows) of carbon that are important on timescales of more than 100 Kyr. Carbon is added to the atmosphere through metamorphic degassing and volcanic activity on land and at mid-ocean ridges. Atmospheric carbon is used in the weathering of silicate minerals in a temperature-sensitive dissolution process; the products of this weathering are carried by rivers to the oceans. Carbonate sedimentation extracts carbon from the oceans and ties it up in the form of limestones. Pelagic limestones deposited in the deep ocean can be subducted and melted. Limestones deposited on continental crust are recycled much more slowly — if they are exposed and weathered, their remains may end up as pelagic carbonates; if they get caught up in a continental collision, they can be metamorphosed, liberating their CO<sub>2</sub>.

# The Carbon Cycle on Earth Now



# Cold Starts?

- As Habitable Zone moves out
  - Can you unfreeze a frozen planet?
  - Will it become suitable for life?
  - If not, HZ will shrink
  - CHZ is smaller

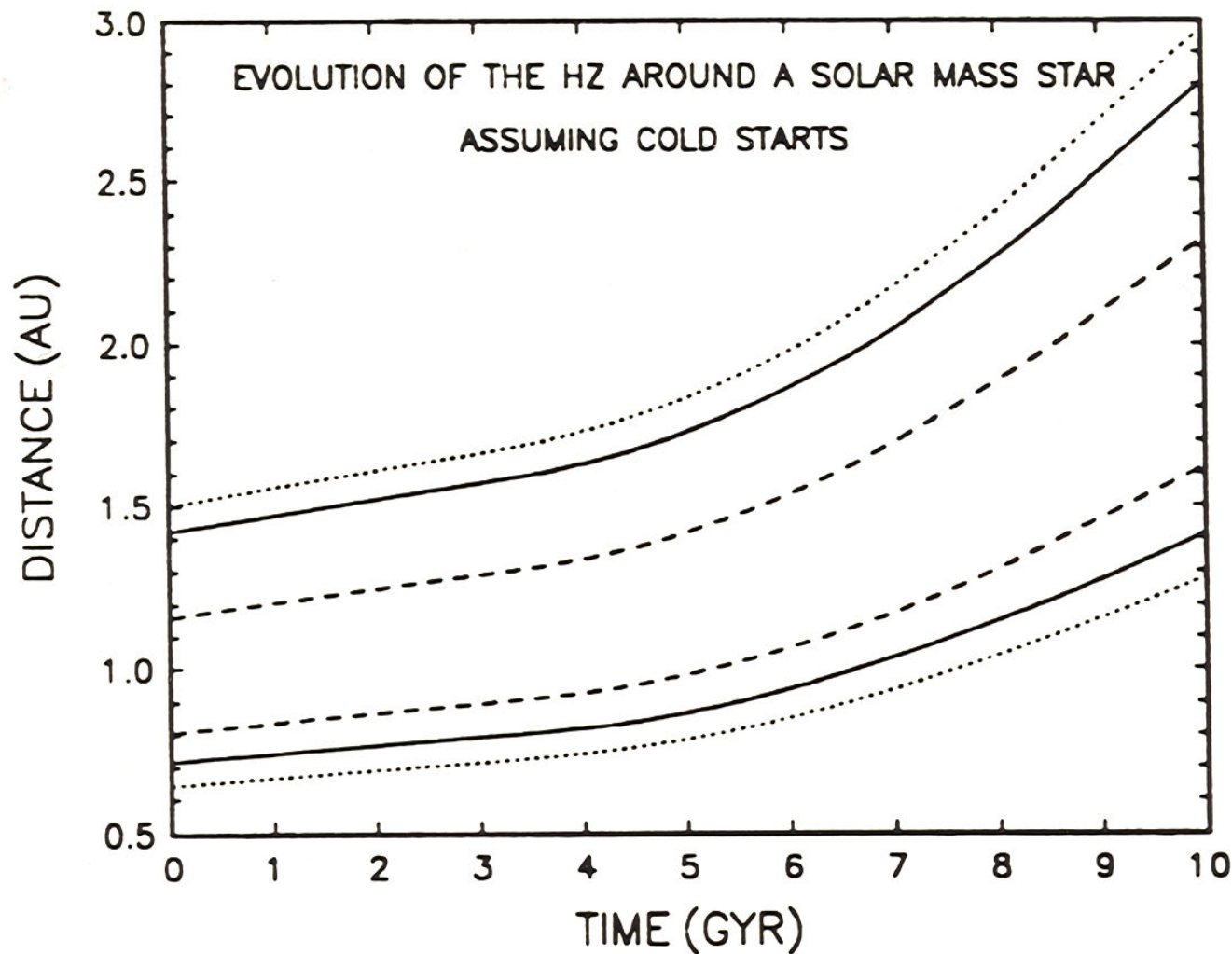


FIG. 2. Evolution of the HZ around a  $1 M_{\odot}$  star assuming that an ice covered planet that was initially beyond the outer HZ boundary can be cold started. The three cases shown are discussed in the text and correspond to the most conservative Case 1 (long dashes), the intermediate Case 2 (solid curves), and the least conservative Case 3 (short dashes).

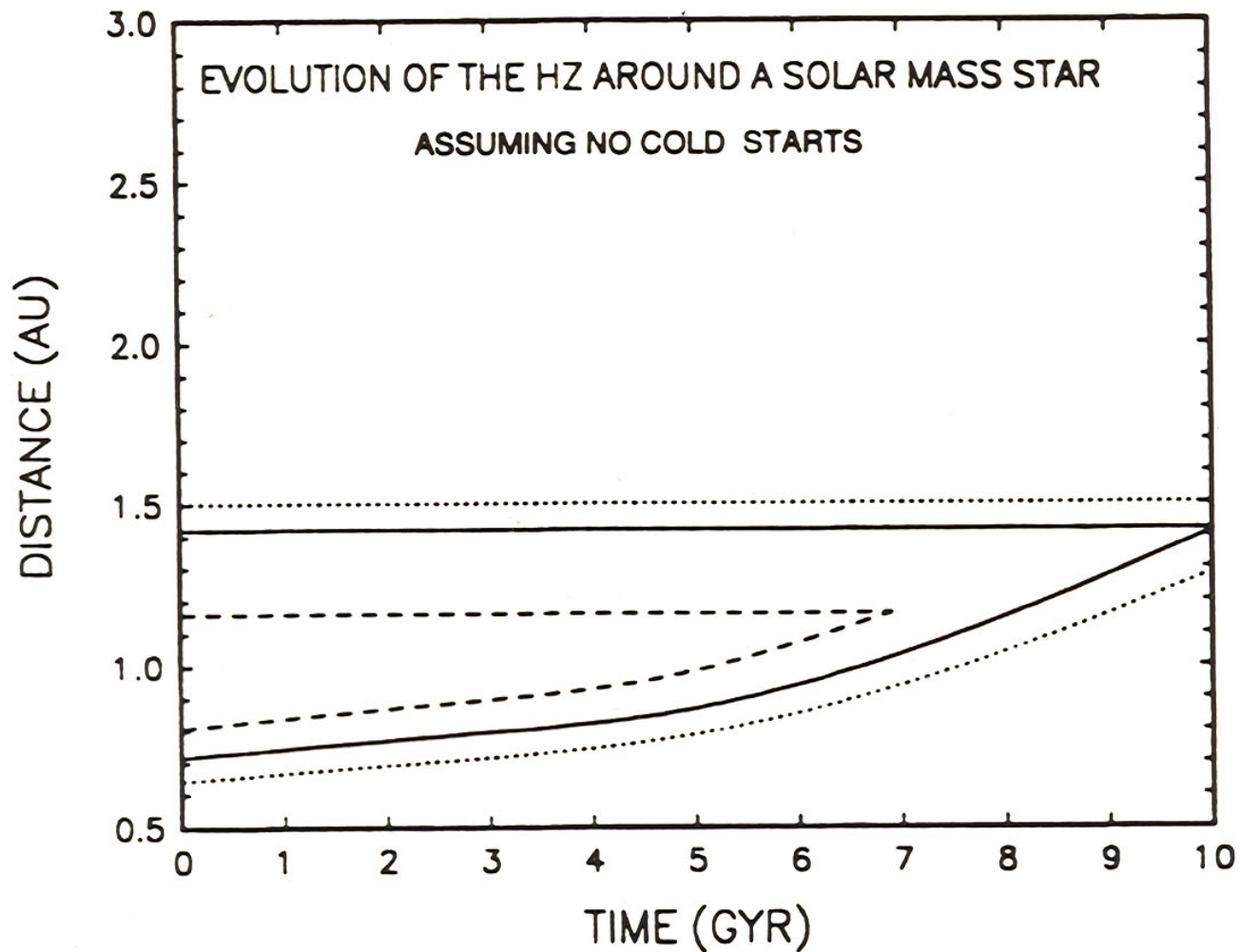


FIG. 3. Evolution of the HZ around a  $1 M_{\odot}$  star assuming that an ice covered planet that was initially beyond the outer HZ boundary cannot be cold started until the stellar flux is greater than the critical greenhouse value. The three cases shown are discussed in the text and correspond to the most conservative Case 1 (long dashes), the intermediate Case 2 (solid curves), and the least conservative Case 3 (short dashes).

# Snowball Earth

Increasing evidence that Earth nearly froze over twice

2.4 billion years ago & 650-800 Myr ago

Climate can have dramatic changes

Apparently - these were ended by volcanic eruptions that put much more CO<sub>2</sub> in atmosphere

# What Else Do We Need?

- So far, we considered liquid water sufficient
- What else?
  - Temperature range
  - Pressure
  - Salt concentration
  - Acidity/alkalinity

# Lower Limit to Temperature?

Some microbes survive for long periods in Antarctic ice

e.g. Lake Vostok - 2.5 miles below glacial ice in Antarctica

Sample obtained in 2013, claims of 3500 species (microbes)

Lake thought to be under ice for last 15 Myr.

But contamination an issue. A supposedly pristine sample was collected in January 2015.

Lower limit is probably about  $-20^{\circ}\text{C}$  (253 K) for active life



# Upper Limit?

We have learned that some microbes can survive in pressurized water at T up to 400 K (120° C)!

Such microbes have special adaptations to protect their heat-sensitive molecules

For complex life, upper limit seems to be ~ 325K

~52° C or 126° F

But is this limit just an accident of evolution on Earth?

## Other Habitable Zones

Microbial Habitable Zone (MHZ)

Fixed by Range of T microbes can withstand

“Animal” Habitable Zone (AHZ)

“Animal” = complex, differentiated, multicellular life

Ward + Brownlee in *Rare Earth* note AHZ much smaller than MHZ

They also argue that parts of our Galaxy unsuitable for animal life

We will consider this last point under  $f_i$

# On the Optimistic Side

## 1. Sub-surface Water?

If you don't need photosynthesis, no need to be on surface

T increases with depth into Earth

⇒ liquid water under “ground”

|       |            |                          |
|-------|------------|--------------------------|
|       | e.g. Mars? | Europa (Moon of Jupiter) |
| HZ    | → 1.5 AU   | 5 AU                     |
| $n_p$ | ~ 2        | ~ 3                      |

## 2. Other Solvents

e.g. Titan (moon of Saturn) has some liquid methane ( $\text{CH}_4$ )

HZ  $\longrightarrow$  10 AU

$n_p$   $\longrightarrow$   $\sim 4$

## 3. Other planetary systems

Jupiter-like planets  $\sim 1$  AU (in HZ)

Life on Moons?

# Other requirements?

Pressure?

Bacteria on deep sea floor withstand up to 1000 atmospheres

But not “animal” life

Not too salty? - halophilic bacteria  
up to 33% salt solution

pH?  $-\text{LOG} [\text{H ions}]$

pH 1

acid

7

normal

$\text{H}_2\text{O}$

pH 14

alkali

Almost all cells regulate pH to 7.7

1 ← microbes → 13

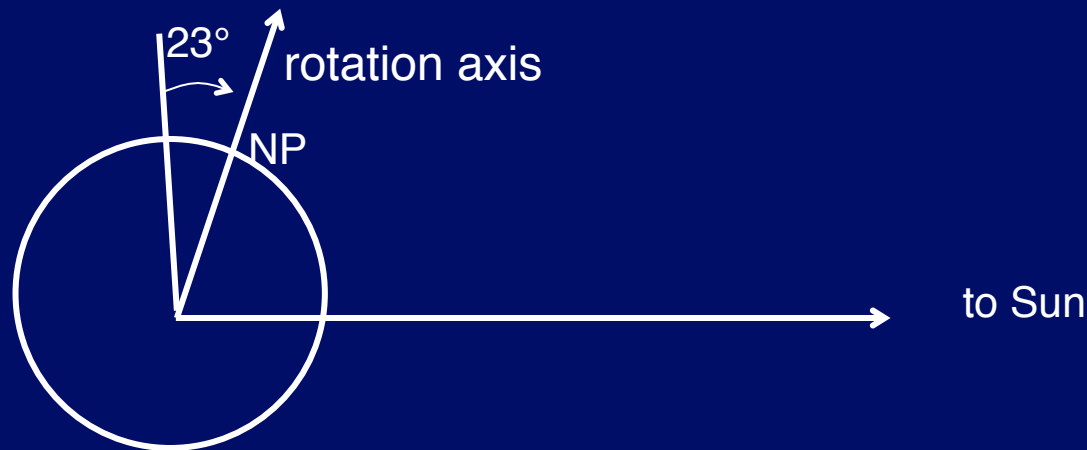
Again, microbes have adapted to just about any environment of Earth

# The Importance of the Moon

The Moon makes the tides bigger than if only the Sun caused tides

May be important in the origin of life

The Moon stabilizes the Earth's obliquity



Varies regularly from 22.1 to 24.5 over 41,000 yrs.

Without the Moon, tugs from other planets  
could make it vary chaotically

Large obliquity could cause snowball Earth

Ward & Brownlee

Only if a large supercontinent at the poles

Williams, Kasting, Caldeira



# Summary

- Complex (animal) life requires “nicer” conditions than microbial life
- Microbes survive in wide range of conditions of T, p, pH, ...
- T set by L, d from star and greenhouse
- HZ and CHZ are important concepts
- Large moon good, but may not be essential
- So far, we assume like solar system