

Agenda for Ast 309N, Sep. 11

- Feedback on 9/6 card and advice on how to answering short essay questions. See card file.
- Solar magnetic field and the solar cycle
- Begin reviewing light: wave-particle duality
- **Quiz 1** is this Thurs., Sep. 13, starting at the **beginning** of class (don't be late!). No need for blue books (you'll answer on sheet provided). Topic: what we have learned about the Sun.

9/11/12

Ast 309N (47760)

1

A few Last Words on Neutrinos

Neutrinos have three properties:

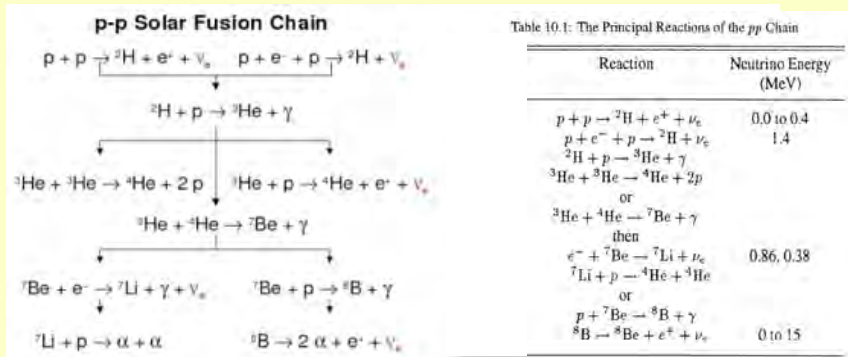
- (1) Type: electron, muon, tau neutrinos – the Sun produces electron neutrinos
- (2) Energy: *not* the same thing as type! A neutrino can have any energy, but those from specific reactions have certain energies
- (3) Neutrinos or anti-neutrinos – Note: the differences are not yet fully understood.

9/11/12

Ast 309N (47760)

2

Neutrinos Made by Fusion in the Sun

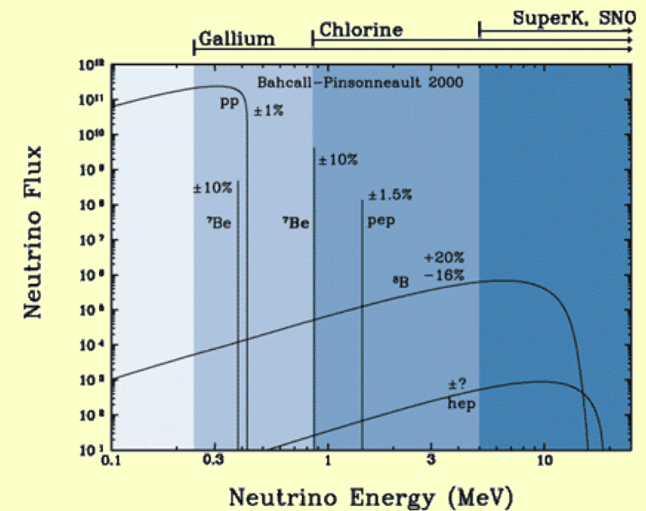


9/11/12

Ast 309N (47760)

3

The Solar Neutrino "Spectrum"



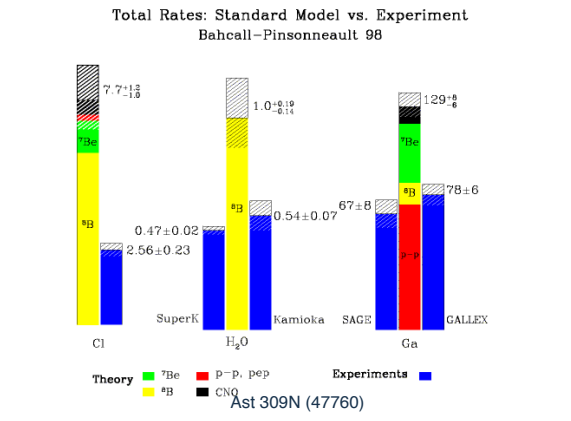
9/11/12

Ast 309N (47760)

4

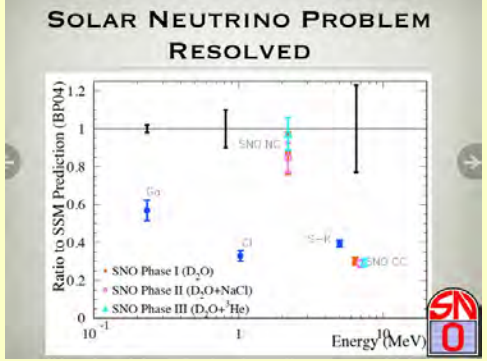
The Solar Neutrino Problem

TOO FEW SOLAR NEUTRINOS DETECTED, COMPARED TO PREDICTIONS FROM MODELS



Sudbury Neutrino Observatory (SNO)

THE NEUTRINOS WERE CHANGING INTO OTHER TYPES ON THEIR WAY FROM THE CORE. THE SNO EXPERIMENT NAILED IT, DETECTING THEM ALL!

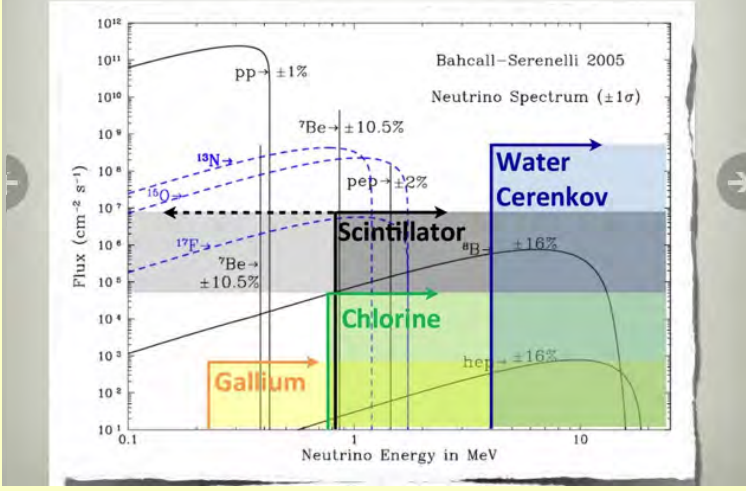


The Solar Neutrino Frontier

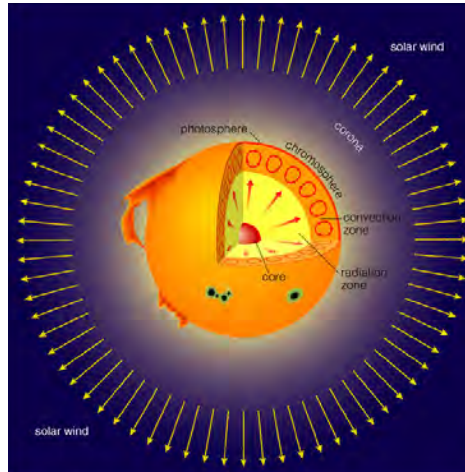
Much of the focus is now on the properties of neutrinos. Current and upcoming experiments:

- (1) SuperK (in Japan): solar, atmospheric, and accelerator-made (KamLAND) neutrinos
- (2) Borexino (in Italy, where the Ga experiments were): sees neutrinos from specific reactions
- (3) SNO+ (SNO-plus): Re-uses the SNO site, with new detector (liquid scintillator); will be a bigger, better version of Borexino

SOLAR ν ENERGY SPECTRA



Outer Layers of the Sun



Solar wind:
flow of charged
particles from
Sun's surface

Corona: outer
layer, $T \sim 10^6$ K;
see in X-rays

Chromosphere:
hotter layer,
 $T \sim 10^4 - 10^5$ K

Photosphere:
visible surface,
 $T \sim 5800$ K

9/11/12

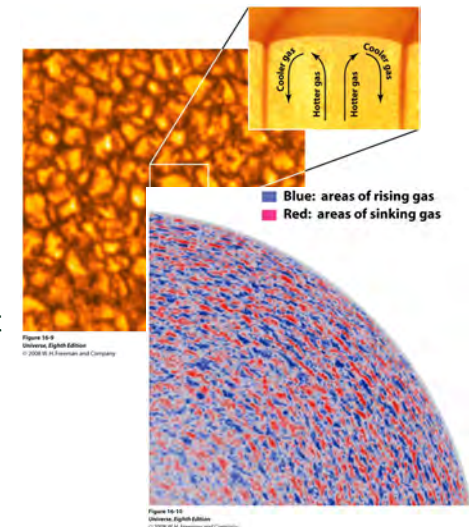
Ast 309N (47760)

9

Evidence of Convection: Solar Surface

Convection brings
energy to the surface
in the form of hot gas

The bright blobs
("granules") on the
photosphere are the
tops of bubbles of hot
gas (convection
'cells') breaking
through the surface

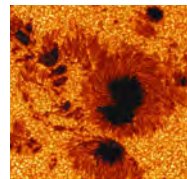
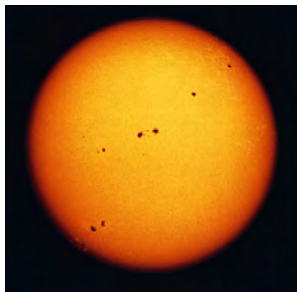


9/11/12

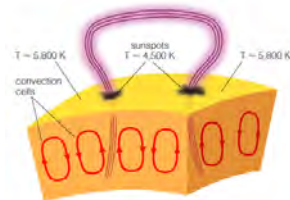
Ast 309N (47760)

10

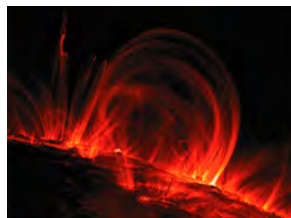
Sunspots



Sunspots appear dark because
they are cooler than adjacent parts
of the Sun's surface. They are the
base of magnetic loops.



← "side view" of magnetic loop



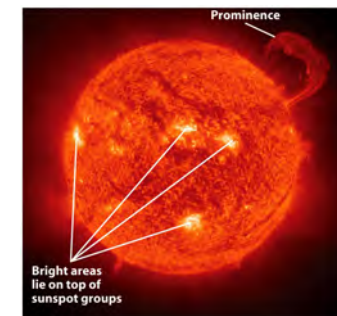
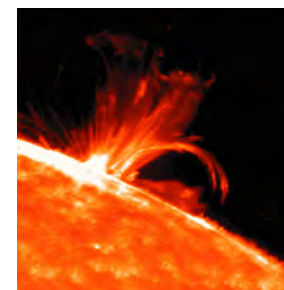
9/11/12

Ast 309N (47760)

11

Solar Prominences and Flares

Solar prominences
are giant loops of hot
gas trapped by
magnetic fields.



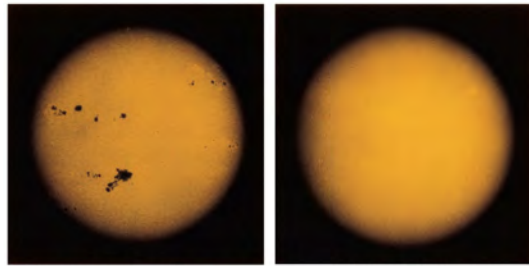
Solar flares happen when a
magnetic loop snaps, sending
a burst of X-rays and charged
particles outwards.

9/11/12

Ast 309N (47760)

12

The Solar Activity Cycle: P = 11 yrs



Near sunspot maximum Near sunspot minimum

Figure 16-18a
Universe, Eighth Edition
© 2008 W. H. Freeman and Company

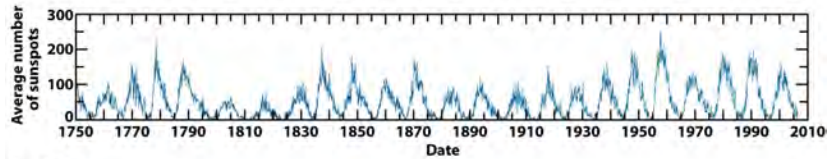
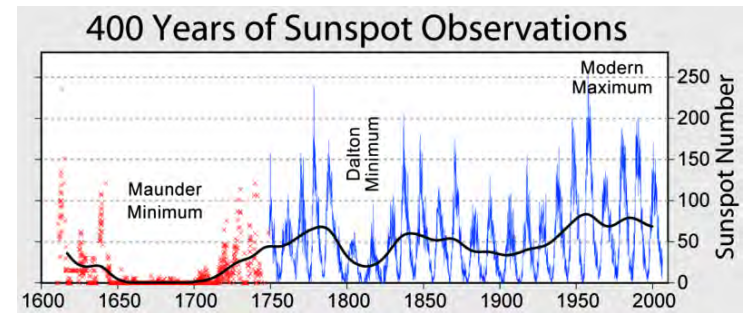


Figure 16-18a
Universe, Eighth Edition
© 2008 W. H. Freeman and Company

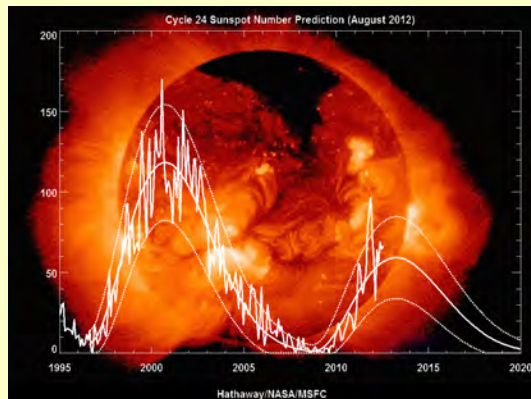
The Solar Activity Cycle, Long-Term

- The Sun is more active during periods called “solar maximum” and less active during “solar minimum.”
- The cycle is approximately 11 years long, but the cycles vary in intensity and sometimes in timing.

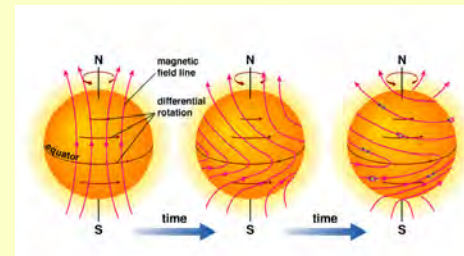


Where are we in the solar cycle?

Approaching solar maximum, but seems to be a wimpy one.



Solar Rotation & Magnetic Field



Copyright © Addison Wesley

Since the Sun rotates differentially, the magnetic field “winds up,” and loops can eventually break out as flares, prominences, etc.

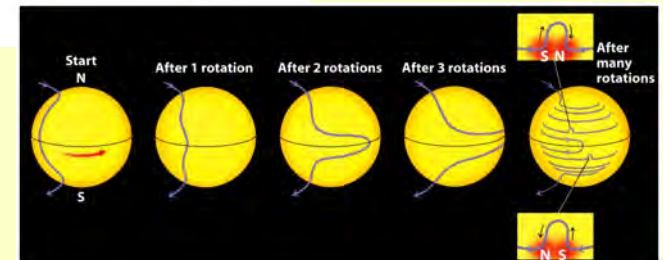


Figure 16-23
Universe, Eighth Edition
© 2008 W. H. Freeman and Company

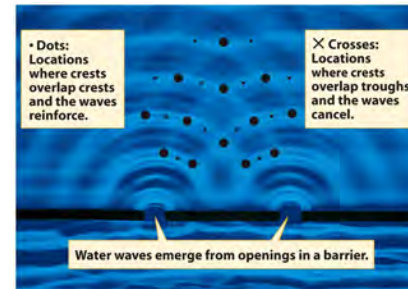
Different Kinds of Light: Colors



Figure 5-3
Universe, Eighth Edition
© 2008 W.H. Freeman and Company

What is “color”? How does red differ from blue light?

In some ways, light acts like a wave



When water waves cross, they interfere with each other; light behaves in a similar way.

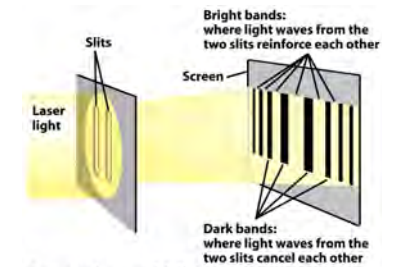
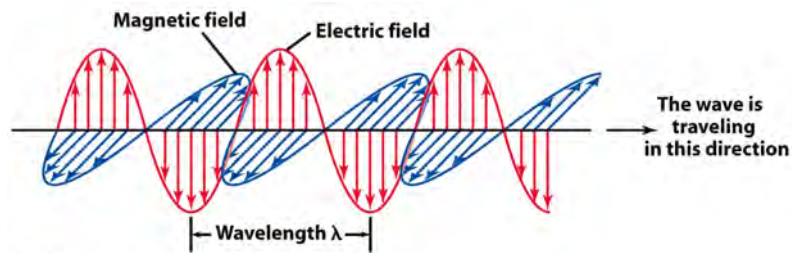


Figure 5-16
Universe, Eighth Edition
© 2008 W.H. Freeman and Company

But a wave of ... what?

- A light wave is a vibration of electric and magnetic fields (a form of energy)
- Light interacts with charged particles (matter!) through these electric and magnetic fields



Properties of Waves

Anatomy of a Wave

Wavelength: the distance between adjacent crests (or troughs)

Amplitude: half the difference in height between a crest and a trough

Frequency: the number of crests that pass through a point (such as the leaf) each second. It is measured in units of hertz (Hz), which are cycles per second

Blink rate = frequency

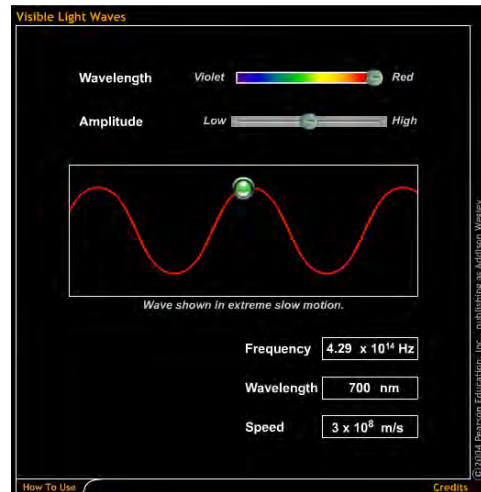
Speed: how fast the pattern of crests and troughs moves forward

Definitions Comparison

How To Use Credits

Light as a Wave

- The speed of a wave is
 $s = f \times \lambda$
- For all kinds of light, the speed s is constant:
 $c = v \times \lambda$
- For higher ν (frequency), you have smaller λ , and vice versa.
- Our eyes interpret light of different λ 's as **different colors**.



9/11/12

Ast 309N (47760)

21

Light Also Acts Like a Particle

- Light energy comes in indivisible packets called photons, which behave like particles
- However, each photon has a specific wavelength and frequency
- The energy of a photon depends on its frequency: $e = h \times \nu$

"wave-particle duality"

9/11/12

Ast 309N (47760)

22

Light as a Stream of Particles

- A beam of light can be thought of as a stream of particles (photons) with various wavelengths.
- Since a photon's energy varies with frequency, it also varies with wavelength:
 $E = h \times \nu = (h \times c) / \lambda$ ["h" = Planck's constant]
- Bluer light has more energy per photon.



9/11/12

Ast 309N (47760)

23

Wavelength, Frequency, and Energy

$$\lambda \times \nu = c$$

$$\lambda = \text{wavelength} \quad \nu = \text{frequency}$$

$$c = 3.00 \times 10^8 \text{ m/s} = \text{speed of light}$$

$$E = h \times \nu = \text{photon energy}$$

$$h = 6.626 \times 10^{-34} \text{ Joule} \cdot \text{s} = \text{photon energy}$$

9/11/12

Ast 309N (47760)

24